TIMSS IEA's Third International Mathematics and Science Study

TIMSS Science Items:

Released Set for Population 2 (Seventh and Eighth Grades)

Overview of TIMSS

TIMSS is a collaborative research project sponsored by the International Association for the Evaluation of Educational Achievement (IEA). In 1994-95, achievement tests in mathematics and science were administered to carefully selected samples of students in classrooms around the world. With more than 40 countries participating, five grades assessed in two school subjects, more than half a million students tested in more than 30 languages, and millions of open-ended responses generated, TIMSS is the largest and most ambitious study of comparative educational achievement ever undertaken.

TIMSS tested and collected contextual information about the schooling of students in the following grade levels:

- Students enrolled in the two adjacent grades that contained the largest proportion of 9-year-olds students – grades 3 and 4 in many countries
- Students enrolled in the two adjacent grades that contained the largest proportion of 13-year-old students – grades 7 and 8 in many countries
- Students in their final year of secondary education. As an additional option, countries could test two special subgroups of these students:
 - Students taking advanced courses in mathematics
 - Students taking advanced courses in physics

The three different groups of TIMSS students listed above are often referred to as Populations 1, 2, and 3, respectively. All countries participated in the testing at Population 2 (grades 7 and 8), which is the core of TIMSS. Table 1 lists the participants that satisfied all of the steps necessary to have their Population 2 science achievement results published in the international report.¹ Countries could choose whether or not to participate in the testing at the other two populations. About 30 countries participated in the testing at Population 1 and about 25 in the testing at Population 3.



Beaton, A.E., Mullis, I.V.S., Martin, M.O., Gonzalez, E.J., Kelly, D.L., and Smith, T.A. (1996). Science Achievement in the Middle School Years: IEA's Third International Mathematics and Science Study (TIMSS). Chestnut Hill, MA: Boston College.

Table 1

TIMSS Part	icipants
Included in the TIMSS Internatio	onal Analyses at Population 2
 Australia Austria Belgium (Flemish) Belgium (French) Bulgaria Canada Colombia Cyprus Czech Republic Denmark England France Germany Greece Hong Kong Hungary Iceland Iran, Islamic Republic Ireland Israel* Japan 	 Korea, Republic of Kuwait* Latvia Lithuania Netherlands New Zealand Norway Portugal Romania Russian Federation Scotland Singapore Slovak Republic Slovenia South Africa Spain Sweden Switzerland Thailand United States

* Participated only at the upper grade.



The success of TIMSS depended on a collaborative effort between the research centers in each country responsible for implementing the project, and the network of centers responsible for managing across-country tasks such as training country representatives in standardized procedures, selecting comparable samples of schools and students, and conducting the various steps required for data processing and analysis. The TIMSS International Study Center, responsible for the international coordination of tasks, is housed in the Center for the Study of Testing, Evaluation, and Educational Policy (CSTEEP) at Boston College.

The TIMSS Science Test

The TIMSS curriculum framework underlying the science tests at all three populations was developed by groups of science educators with input from the TIMSS National Research Coordinators (NRCs).² The **content** aspect of the framework represents the subject matter content of school science. The **performance expectations** aspect of the framework describes, in a non-hierarchical way, the many kinds of performances or behaviors that might be expected of students in school science. Working within the science curriculum framework, science test specifications were developed for Population 2 that included items representing a wide range of science topics and eliciting a range of skills from the students.

The tests were developed through an international consensus involving input from experts in science and measurement specialists.³ The TIMSS Subject Matter Advisory Committee, which included distinguished scholars from 10 countries, ensured that the test reflected current thinking and priorities within the field of science. The items underwent an iterative development and review process with one pilot testing effort involving 43 countries. Every effort was made to help ensure that the tests represented the curricula of the participating countries and that the items did not exhibit any bias towards or against particular countries, including modifying specifications in accordance with data from the curriculum analysis component, obtaining ratings of the items by subject matter specialists within the participating countries, and conducting thorough statistical item analysis of data collected in the pilot testing. The final forms of the test were endorsed by the NRCs of all the participating countries. The resulting test for the Population 2 students (seventh and eighth grades in many countries) contained 135 science items representing a range of science topics and skills.

Approximately one-fourth of the TIMSS items were in the free-response format, which required students to generate and write their own answers. Designed to represent approximately one-third of students' response time, some free-response questions asked for short answers, while others called for extended responses and required students to show their work. The remaining questions used a multiple-choice format. The distribution of items across content areas (as reported in the international reports) and performance expectations, as well as by item format, is presented in Table 2.

² The complete TIMSS curriculum frameworks can be found in Robitaille, D.F. et al. (1993). *TIMSS Monograph No. 1: Curriculum Frameworks for Mathematics and Science*. Vancouver, B.C.: Pacific Educational Press.

³ Please see Garden, R.A. (1996), "Development of qÚ/TIMSS Achievement Items" in D.F. Robitaille and R.A. Garden (Eds.), *TIMSS Monograph No. 2: Research Questions and Study Design*. Vancouver, B.C. Pacific Education Press; and Garden, R.A. and Orpwood, G. (1996). "Development of the TIMSS Achievement Test" in M.O. Martin and D.L. Kelly (Eds.), *Third International Mathematics and Science Study Technical Report, Volume I: Design and Development*. Chestnut Hill, MA: Boston College.

Table 2

Distribution of Science Items by Content Reporting Category and Performance Expectation¹ - Population 2

Content Category	Number of Items	Number of Multiple- Choice Items	Number of Free- Response Items ²
Earth Science	22 (11)	17 (6)	5 (5)
Life Science	40 (27)	31 (18)	9 (9)
Physics	40 (25)	28 (13)	12 (12)
Chemistry	19 (14)	15 (10)	4 (4)
Environmental Issues and the Nature of Science	14 (10)	11 (7)	3 (3)
Total	135 (87)	102 (54)	33 (33)

Performance Expectation	Number of Items	Number of Multiple- Choice Items	Number of Free- Response Items ²
Understanding Simple Information	55 (28)	53 (26)	2 (2)
Understanding Complex Information	39 (26)	29 (16)	10 (10)
Theorizing, Analyzing, and Solving Problems	28 (24)	9 (5)	19 (19)
Using Tools, Routine Procedures, and Science Processes	8 (4)	8 (4)	0 (0)
Investigating the Natural World	5 (5)	3 (3)	2 (2)

¹Figure in parentheses refers to the number of items in the released item set and provided in this volume. ²Free-Response Items include both short-answer and extended-response types.

SOURCE: IEA Third International Mathematics and Science Study (TIMSS), 1994-95.



To ensure broad subject matter coverage without overburdening individual students, TIMSS used a rotated design that included both the mathematics and science items. In accordance with the design, the mathematics and science items were assembled in 26 different clusters — labeled A through Z. The clusters were assigned to eight different booklets in accordance with the rotated design so that representative samples of students responded to each cluster.⁴ Each student completed one 90-minute test booklet containing both mathematics and science items.

Item Release Policy

In accordance with IEA policy, TIMSS has kept about one-third of the TIMSS items secure for possible future use in measuring international trends in mathematics and science achievement. For Population 2, the secure items are in clusters labeled A through H. All remaining items (in clusters I through Z) are available for general use. To facilitate this use, the released TIMSS items for Population 2 (seventh and eighth grades) have been replicated in their entirety in this science volume and in the companion mathematics volume. As shown in Table 2, this volume contains 87 science items, including all of the free-response questions. To provide a unique identifier for each item, the TIMSS cluster and item number is shown in the black box on the right hand side of each page.

While the purpose of this volume is to encourage the use of TIMSS and TIMSS items, please note the IEA copyright; appropriate references to the IEA and TIMSS should be provided in your use of these items.

Item Documentation and Item Results

The TIMSS tests were prepared in English and translated into 30 additional languages. Each item is reproduced for this volume exactly as it was presented to each of the TIMSS countries. In translating the tests or making adaptations for cultural purposes, every effort was made to ensure that the meaning and difficulty of items did not change. This process required an enormous effort by the national centers, with many checks made along the way.⁵

Across the bottom of each item, there is documentation about the item, including the subject assessed and the classification of the item by content category and performance expectation. If the item is a two-part item, the documentation for Part A is shown on the first page and the documentation for Part B is shown on the following page.

⁴ The TIMSS test design is fully documented in Adams, R. and Gonzalez, E. (1996). "Design of the TIMSS Achievement Instruments" in D.F. Robitaille and R.A. Garden (Eds.), *TIMSS Monograph No. 2: Research Questions and Study Design*. Vancouver, B.C.: Pacific Education Press; and Adams, R. and Gonzalez, E. (1996). "TIMSS Test Design" in M.O. Martin and D.L. Kelly (Eds.), *Third International Mathematics and Science Study Technical Report, Volume I: Design and Development*. Chestnut Hill, MA: Boston College.

⁵ More details about the translation verification procedures can be found in Mullis, I.V.S., Kelly, D.L., and Haley, K. (1996). "Translation Verification Procedures" in M.O. Martin and I.V.S. Mullis (Eds.), *Third International Mathematics and Science Study: Quality Assurance in Data Collection*. Chestnut Hill, MA: Boston College; and Maxwell, B. (1996). "Translation and Cultural Adaptation of the TIMSS Instruments" in M.O. Martin and D.L. Kelly (Eds.), *Third International Mathematics and Science Study Technical Report, Volume I*. Chestnut Hill, MA: Boston College.



Subject. All of the items in this volume are science items. The mathematics items are provided in a companion volume, *TIMSS Mathematics Items: Released Set for Population 2 (Seventh and Eighth Grades)*.

Key. For multiple-choice items, the key for the correct answer is provided. For freeresponse questions, the categories of responses and their codes are shown on the page following the item. In scoring the TIMSS free-response questions, TIMSS utilized two-digit codes with rubrics specific to each item. The first digit designates the correctness level of the response. The first digit is usually a "1" designating a correct response, a "7" indicating an incorrect response, or a "9" for non-response. Sometimes, however, fully correct responses are differentiated from partially correct responses. In these instances, the fully correct responses are designated by a "2" (or in one instance by a "3") and the partially correct responses by a "1." The second digit, combined with the first digit, represents a diagnostic code used to identify specific types of approaches, strategies, or common errors and misconceptions.

Content Category. The science items were reported according to five content areas.

- ► Earth Science
- ► Life Science
- ► Physics
- ► Chemistry
- ▶ Environmental Issues and the Nature of Science

Table 3 indicates which items have been classified into each of the five content areas.

Performance Expectation. Items were classified into the following performance expectations.

- ► Understanding Simple Information
- ► Understanding Complex Information
- ▶ Theorizing, Analyzing, and Solving Problems
- ▶ Using Tools, Routine Procedures, and Science Processes
- ▶ Investigating the Natural World

Percent of Students Responding Correctly. The percent of students responding correctly to the item reflects the international average across the countries participating in TIMSS at each grade tested. That is, first the percentage of students responding correctly to the item was calculated for each country. Next, an average was calculated across countries. For the upper grade (eighth grade in many countries), this average was calculated across 41 countries (see Table 1). For the lower grade (seventh grade in many countries), the average is based on 39 countries. For items using a partial credit scoring scheme, the percentages given are for students responding with fully correct answers.

International Difficulty Index. This statistic reflects the difficulty of the item as estimated from item response theory scaling (IRT). Since the TIMSS scale was developed based on the performance of students at both grades in all countries, the international scale values apply to both grades and to all countries. The higher the index, the more difficult the item.

Table 3

Item Listing by Science Content Area

Earth Science

- I17 Energy for Earth's water cycle.
- J01 Description of Earth's surface.
- K15 Formation of fossil fuels.
- O12 Gases in air. O14 Sun and moon.
- P03 Life on planet Athena.
- Q11 Daylight and darkness.
- Q16 Light from star.
- R04 Ozone layer.
- W01A Farming on plain (a).
- W01B Farming on plain (b).
- W02 Diagram of Earth's water cycle.

Life Science

I10 Fruits and vegetables in diet 111 Insect features 114 Arm represents what simple machine? 119 Statement consistent with data in table. J02 Species on Earth. J07 Warm-blooded vs. cold-blooded animals. J09 Tree rings K11 Interdependence among aquatic organisms. Would this reduce population? K12 K16 Which made with help of bacteria? K18 Chloroplasts in cells. L02 Green algae in ocean. 1.03 Skull of animal. L05 Reason for bird singing. L06 Snakes and birds on a cold day. M11 Food web N02 Meal with most nutrients. N04 Decaying fish and plant growth. N06 Most basic unit of living things. O16 Thirsty on a hot day. Jose's influenza. 017 P04 Animal hibernation. P06 Digestive substance in the mouth. Q17 Advantage of two eyes. R03 New species in area. Heart rate changes. X01 X02A Aquarium (a). X02B Aquarium (b).

Physics

Which spoon will feel hottest? 116 Solar radiation and sunburn. .105 K10 Show air exists. K13 Light bulb in circuit. Water loss due to evaporation. K14 Falling apple. K17 Forces on rod. L01 L04 Two machines. L07 Communication in space. M12 Using ammeter. M14 Pencil in the mirror. N08 Balance on seesaw. N10 Watering can. 010 Magnet. O13 Ball in groove. Distance versus time graph. P01 P02 Flashlight shining on wall. Heating water with balloon. P05 Jim and Sandy's flashlights. Q12 Q13 Lid on jar. Q18 Melting ice cubes. R01 Light striking mirror. Why does shirt look blue? R02 Y01 Energy in a lamp. Warming snowballs Y02

Chemistry

- J03 Molecules, atoms, and cells.
- J04 Example of chemical reaction.
- J06 What happens to atoms after death?
- J08 Gas causing flame.
- M10 Which is not a mixture?
- M13 Oil burning.
- N07 Glass over candle.
- N09 Filtration.
- O11 Chemical change
- O15 Atom loses electron.
- Q14 Heated iron and sulphur.
- Q15 Chemical change.
- R05 CO2 fire extinguisher.
- Z01A Painting the bridge (a).
- Z01B Painting the bridge (b).Z01C Painting the bridge (b).

Environmental Issues and the Nature of Science

- I12 Comparing trials to test idea.
- I13 Thermometer to measure body temperature.
- I15 Type of statement made.
- I18 Juanita's experiment.K19 How computers help.
- N01 Plant/mineral experiment.
- N03 Liquid evaporation experiment.
- N05 Acid rain.
- P07 Replication of measurements.
- Z02A Water allocation.
- Z02B Water allocation.



For More Information About TIMSS

For more details about the TIMSS results and procedures, please see the following reports:

Mathematics Achievement in the Middle School Years: IEA's Third International Mathematics and Science Study. Beaton, A.E., Mullis, I.V.S., Martin, M.O., Gonzalez, E.J., Kelly, D.L., and Smith, T.A. Chestnut Hill, MA: Boston College, 1996.

Science Achievement in the Middle School Years: IEA's Third International Mathematics and Science Study. Beaton, A.E., Martin, M.O., Mullis, I.V.S., Gonzalez, E.J., Smith, T.A., and Kelly, D.L. Chestnut Hill, MA: Boston College, 1996.

Mathematics Achievement in the Primary School Years: IEA's Third International Mathematics and Science Study. Mullis, I.V.S., Martin, M.O., Beaton, A.E., Gonzalez, E.J., Kelly, D.L., and Smith, T.A. Chestnut Hill, MA: Boston College, 1997.

Science Achievement in the Primary School Years: IEA's Third International Mathematics and Science Study. Martin, M.O., Mullis, I.V.S., Beaton, A.E., Gonzalez, E.J., Smith, T.A., and Kelly, D.L. Chestnut Hill, MA: Boston College, 1997.

Third International Mathematics and Science Study Technical Report, Volume I: Design and Development. Martin, M.O. and Kelly, D.L., Eds. Chestnut Hill, MA: Boston College, 1996.

Third International Mathematics and Science Study: Quality Assurance in Data Collection. Martin, M.O. and Mullis, I.V.S., Eds. Chestnut Hill, MA: Boston College, 1996.

These reports can be ordered from the International Study Center at Boston College.

- ► To FAX Order: +1(617)552-8419
- ► To Phone Order: +1(617)552-4521
- ► To E-mail Order: timss@bc.edu

TIMSS reports and this released item set are also available on the World Wide Web:

http://wwwcsteep.bc.edu/timss

Released Science Items Population 2





Subject	Item Key	Content Category	Performance	Respondin	g Correctly	Difficulty
			Expectation	Upper Grade	Lower Grade	Index
Science	С	Life Science	Understanding Simple Information	74%	72%	433





Investigating the Natural

World

Environmental Issues and

the Nature of Science

Science

D

I-12

Lower Grade

30%

677

37%



Using Tools, Routine

Processes

Procedures, and Science

61%

54%

525

Environmental Issues and

the Nature of Science

Science

А



Theorizing, Analyzing,

and Solving Problems

54%

51%

561

Science

D

Life Science

I-14

I-15

I15. Maria collected the gas given off by a glowing piece of charcoal. The gas was then bubbled through a small amount of colorless limewater. Part of Maria's report stated, "After the gas was put into the jar, the limewater gradually changed to a milky white color." This statement is

an observation

Α.

B.

a conclusion

- a generalization
- an assumption of the investigation D.

FUX CUL

a hypothesis E.

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Minit Calif exc

ress

Subject	Item Key	Content Category	Performance	Internationa Percent of Respondin	al Average Students g Correctly	International Difficulty
			Expectation	Upper Grade	Lower Grade	Index
Science	A	Environmental Issues and the Nature of Science	Using Tools, Routine Procedures, and Science Processes	50%	43%	594



Understanding Simple

Information

83%

80%

341

Science

А

Physics





Subject	Item Key	Content Category	Performance	Responding	g Correctly	Difficulty
			Expectation	Upper Grade	Lower Grade	Index
Science	next page	Environmental Issues and the Nature of Science	Theorizing, Analyzing, and Solving Problems	38%	30%	678

I-18 Coding Guide



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Note: "Incorrect" refers to responses in terms of the conclusions from Juanita's experiment. Thus, some correct statements can be coded 79 (e.g., Moist grains of corn germinate better than dry grains of corn.) Also, simply circling or selecting statement #1 is coded 70 and circling or selecting statement #2 as code 71.

Nin ninnes

JIPUS

Code	Response
Correct	Response
10	Light is not required for moist corn to germinate.
	Examples: Corn can germinate also in the dark.
	Corn needs water and not light to germinate.
	Moist grains of corn can germinate in the light and the dark.
19	Other correct.
Incorrec	t Response
70	Light is required for germination.
	Example: All seeds need light.
71	Darkness is required for germination.
	Example: Seeds germinate in the dark.
79	Other incorrect.
Nonres	oonse
90	Crossed out/erased, illegible, or impossible to interpret.
99	BLANK



Investigating the Natural

World

53%

45%

582

Science

А

Life Science

J1. Which BEST describes the surface of the Earth over billions of years?

Β.

A. A flat surface is gradually pushed up into higher and higher mountains until the Earth is covered with mountains.

High mountains gradually wear down until most of the Earth is at sea level.

High mountains gradually wear down as new mountains are continuously being formed, over and over again.

D. High mountains and flat plains stay side by side for billions of years with little change.

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Subject	ltem Key	Content Category	Performance	Internationa Percent of Respondin	al Average Students g Correctly	International Difficulty
			Expectation	Upper Grade	Lower Grade	Index
Science	С	Earth Science	Understanding Complex Information	41%	36%	651

In the con



Information

66%

61%

505

Science

А

Life Science

J-2



Understanding Simple

Information

32%

21%

726

next

page

Chemistry

Science

J-3 Coding Guide



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	This tenne ut en front	
Code	Response	
	Kesponse	
	et Posponso	
70	Cells - Atoms - Molecules	
71	Molecules - Atoms - Cells.	
72	Molecules - Cells - Atoms.	
73	Atoms - Molecules - Cells.	
74	Atoms - Cells - Molecules.	
79	Other incorrect.	
Nonres	ponse	
90	Crossed out/erased, illegible, or impossible to interpret.	
99	BLANK	



Exposition	Upper Grade	Lower Grade
Understanding Simple Information	47%	35%

Science

С

Chemistry

635



J6. Animals are made up of many atoms. What happens to the atoms after an animal has died? The atoms stop moving. Α. В. The atoms recycle back into the environment. C. The atoms split into simpler parts and then combine to form other atoms. The atoms no longer exist once the animal has decomposed. mercial purposes mercial press frances mithout on from the permission nistennine karcomme Reproduced from TIMSS Population 2 Item Pool. Copyright © 1994 by IEA, The Hague International Average Percent of Students International **Responding Correctly** Subject Item Key **Content Category** Performance Difficulty Expectation Index Upper Grade Lower Grade

Understanding Complex

Information

26%

22%

761

Science

В

Chemistry

- J7. How are warm-blooded animals different from cold-blooded animals?
 - Warm-blooded animals have a higher metabolism in warm weather. A.
 - Warm-blooded animals are more aggressive in captivity.

Β.

C

D.

- Warm-blooded animals always have a higher blood temperature.
- Warm-blooded animals normally maintain a fairly constant internal temperature at all air temperatures.

withcut

hist conn

Warm-blooded animals are found only in warm climates. E.

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Introve expression from

Subject	Item Key	Content Category	Performance	Internationa Percent of Respondin	al Average Students g Correctly	International Difficulty
			Expectation	Upper Grade	Lower Grade	Index
Science	D	Life Science	Understanding Simple Information	52%	45%	600

19

J-7





page

J-9 Coding Guide

J9. How ce	ould you find out how old a tree is after it is cut?
	A2 201
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16	
0	
· O`x	
X	
$\langle O \rangle$	
)	
	ALL ALL S
Code	Response
Correct	Response
10	Response includes counting rings in the trunk and explicitly
	<i>Example: Count the rings in the trunk, one per year</i>
11	Response includes counting the rings in the trunk, but there is
19	no mention of one ring per year.
Incorrec	t Response
70	Mentions looking at other parts of trees, not the trunk, including
70	roots, bark, etc.
/9	Example: By asking someone who studies trees.
Nonres	ponse
90	Crossed out/erased illegible or impossible to interpret
	crossed out crused, megicie, or impossible to interpret.



K-10 Coding Guide

UN CTECTER

K10. Air is colorless, odorless, and tasteless. Describe one way that air can be shown to exist.

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Code Response Correct Response Image: A provide the set of	je ^č	ee hotoeuse hotoeuse hotoeuse hotoeuse	0 585 55 (F.A
Correct Response 10 Mentions that you can feel or see effects of air movement. Examples: Wind, flags blowing, waving arms, spreading smell. 11 Mentions that (light) things fall slowly. Example: Air resistance makes things fall slowly. 12 Refers to the fact that air can be weighed. 13 Mentions that balloons or tires, etc. can be filled with air. 14 Refers to air pressure. Example: Barometers show that air exists. 15 Refers to being able to 'see' air. Example: You can see air bubbles in water. 19 Other correct. Incorrect Response 70 We can breathe air. Refers only to the need of oxygen or air for life and other processes. Examples: All living things need air/oxygen Candles extinguish without air. We would die if there was no air. 72 Refers to seeing water vapor. Example: You can see water "vapor" when breathing out [on cold days or on a mirror or glass]. 76 Merely repeats information in the stem. 79 Other incorrect. Nonresponse 90 90 Crossed out/erased, illegible, or impossible to interpret. 99 BLANK	Code	Response	()
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K11. The diagram below shows an example of interdependence among aquatic organisms. During the day the organisms either use up or give off (a) or (b) as shown by the arrows. K-11 Floatingb water plant a` Small water animals h Water plant with roots Choose the right answer for (a) and (b) from the alternatives given. AL JAK (a) is oxygen and (b) is carbon dioxide. A. (a) is oxygen and (b) is carbohydrate. B. (a) is nitrogen and (b) is carbon dioxide. C. (a) is carbon dioxide and (b) is oxygen. D. (a) is carbon dioxide and (b) is carbohydrate. E. Reproduced from TIMSS Population 2 Item Pool. Copyright © 1994 by IEA, The Hague Т International Average

Subject	ltem Key	Content Category	Performance Expectation	Percent of Students Responding Correctly		International Difficulty
				Upper Grade	Lower Grade	Index
Science	А	Life Science	Understanding Complex Information	59%	51%	551

K12. The male insects in a population are treated to prevent sperm production. Would this reduce this insect population?

No, because the insects would still mate. A.

No, because it would not change the offspring mutation rate.

Yes, because it would sharply decrease the reproduction rate.

Yes, because the males would die.

AISTENNING FOR COMME

Β.

C.

D.

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Subject	Item Key	Content Category	Performance Expectation	International Average Percent of Students Responding Correctly		International Difficulty
				Upper Grade	Lower Grade	Index
Science	С	Life Science	Theorizing, Analyzing, and Solving Problems	55%	50%	570

K-12



Subject	ltem Key	Content Category	Performance Expectation	Percent of Students Responding Correctly		International Difficulty
				Upper Grade	Lower Grade	Index
Science	В	Physics	Understanding Complex Information	78%	69%	429










Understanding Simple

Information

54%

50%

557

Science

А

Life Science

K-18



K-19 Coding Guide

K19. Write down one example of how computers help people do their work.

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19 Othe Exam It we	ponse refers to any combination of two or more of codes 3.
ı —	er correct: mples: The computer does no mistakes. orks faster.
Incorrect Re	esponse
70 Play	ing games such as Nintendo.
71 Vag	ue references to "everything" or some similar expression.
76 Mer	
79 Othe	ely repeats information in the stem.
Nonrespon	er incorrect.
90 Cro	er incorrect.
99 BLA	er incorrect. Se assed out/erased, illegible, or impossible to interpret



Information

Which BEST explains why green marine algae are most often restricted to the L2. top 100 meters of the ocean?

They have no roots to anchor them to the ocean floor.

They can live only where there is light.

nistennine korconne

A.

В.

C.

D.

The pressure is too great for them to survive below 100 meters.

If the algae lived below 100 meters they would be eaten by animals.

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Subject	ltem Key	Content Category	Performance	Internationa Percent of Respondin	al Average Students g Correctly	International Difficulty
			Expectation	Upper Grade	Lower Grade	Index
Science	В	Life Science	Understanding Complex Information	53%	47%	574

L-2

36

- L3. A girl found the skull of an animal. She did not know what the animal was but she was sure that it preyed on other animals for its food. What clue led to this conclusion?
 - The eye sockets faced sideways.

A.

C.

- Β. The skull was much longer than it was wide.
 - There was a projecting ridge along the top of the skull.
- Four of the teeth were long and pointed. D.

the connor

The jaws could move sideways as well as up and down. E.

HUNSSON FROM

Subject	ltem Key	Content Category	Performance	Internationa Percent of Respondin	al Average Students g Correctly	International Difficulty
			Expectation	Upper Grade	Lower Grade	Index
Science	D	Life Science	Understanding Complex Information	71%	67%	471

L-3

L4. Machine A and Machine B are each used to clear a field. The table shows how large an area each cleared in 1 hour and how much gasoline each used.

Machine A	Area of field cleared in 1 hour	Gasoline used in 1 hour
Machine A		
	2 hectares	3/4 liter
Machine B	1 hectare	1/2 liter
		J. 65
Which machi	ne is more efficient in convert	ing the energy in gasoline to work?
Explain your	answer.	×Y
	itsitemmer for with per	cial Presition

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Subject	Item Key	Content Category	Performance	Internationa Percent of Respondin	al Average Students g Correctly	International Difficulty
			Expectation	Upper Grade	Lower Grade	Index
Science	next page	Physics	Theorizing, Analyzing, and Solving Problems	36%	29%	688

38

L-4 Coding Guide

Machine A and Machine B are each used to clear a field. The table shows how large L4. an area each cleared in 1 hour and how much gasoline each used.

		Area of field cleared in 1 hour	Gasoline used in 1 hour
	Machine A	2 hectares	3/4 liter
	Machine B	1 hectare	1/2 liter
	1		
•	Which mach	ine is more efficient in converting	the energy in gasoline to work? Explain
	your answer	c 🚺 🔪	

Reportion LMSS Population 2 Item Pool. Copyright 8 1994 by IEA. The Hague Reporting the second sec	Which ma	chine is more efficient in converting the energy in gasoline to work? Explain	
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Code Response 10 A. Because it uses less gas per hectare. 11 A. Because it uses less gas per hectare. 11 A. Because 3/8 < 1/2, OR a similar expression. 19 A. Other correct. Incorrect Response 70 B. With an explanation. 71 B. No explanation. 72 States that A and B are the same, with an explanation. 73 States that A and B are the same. No explanation. 74 A. Mentions that it clears the most area per hour. 75 A. Other wrong explanation. 76 A. No explanation. 77 A. Other wrong explanation. 79 Other incorrect. Nonresponse 90 90 BLANK	Reproduced from	TIMSS Population 2 Item Pool. Copyright © 1994 by IEA, The Hague	
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74 A. Mentions that it clears the most area per hour. 75 A. Other wrong explanation. 76 A. No explanation. 79 Other incorrect. Nonresponse 90 Crossed out/erased, illegible, or impossible to interpret. 99 BLANK	73	States that A and B are the same. No explanation.	
75 A. Other wrong explanation. 76 A. No explanation. 79 Other incorrect. Nonresponse 90 Crossed out/erased, illegible, or impossible to interpret. 99 BLANK	74	A. Mentions that it clears the most area per hour.	
70 A. No explanation. 79 Other incorrect. Nonresponse 90 90 Crossed out/erased, illegible, or impossible to interpret. 99 BLANK	75	A. Utner wrong explanation.	
Nonresponse 90 Crossed out/erased, illegible, or impossible to interpret. 99 BLANK	70	Other incorrect	
90 Crossed out/erased, illegible, or impossible to interpret. 99 BLANK			
90Crossed out/erased, illegible, or impossible to interpret.99BLANK	Nonres	ponse	
99 BLANK	90	Crossed out/erased, illegible, or impossible to interpret.	
· · · · · · · · · · · · · · · · · · ·	99	BLANK	



Information

40

L6. On cold days, snakes usually lie very still and eat little or nothing, while birds usually move around and eat a lot of food. Which statement best explains this?

A.

B.

- Both animals are cold-blooded, but without feathers to keep warm, snakes get too cold to move.
- Unlike birds, snakes are warm-blooded; they must hibernate during cold weather.
- Unlike snakes, birds are cold-blooded; they are less affected by the cold than snakes.

D. Unlike snakes, birds are warm-blooded; they must eat food to maintain a constant temperature.

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HUUL CAROMI

Subject	ltem Key	Content Category	Performance	Internationa Percent of Respondin	al Average Students g Correctly	International Difficulty
			Expectation	Upper Grade	Lower Grade	Index
Science	D	Life Science	Theorizing, Analyzing, and Solving Problems	54%	50%	571

his con

- L7. The crews of two boats at sea can communicate with each other by shouting. Why is it impossible for the crews of two spaceships a similar distance apart in space to do this?
 - The sound is reflected more in space.

A.

B

D.

The pressure is too high inside the spaceships.

the conner

- The spaceships are traveling faster than sound.
- There is no air in space for the sound to travel through.

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MILLOUTEX

Subject	Item Key	Content Category	Performance	Internationa Percent of Respondin	al Average Students g Correctly	International Difficulty
			Expectation	Upper Grade	Lower Grade	Index
Science	D	Physics	Understanding Complex Information	70%	67%	473





M-11 Coding Guide



Code	Response					
Correct	Correct Response					
10	2-3-4-1 (see diagram) or appropriate names of plants/animals					
11	Three of the numbers are correct; one is missing.					
Incorrec	t Response					
70	No number is correctly entered.					
71	Only 2 is correct.					
	Only 3 is correct. Only 4 is correct.					
72	2 and 3 only are correct. 2 and 4 only are correct. 2 and 1 only are correct. 3 and 4 only are correct. 3 and 1 only are correct. 4 and 1 only are correct.					
79	Other incorrect.					
Nonres	Donse					
90	Crossed out/erased, illegible, or impossible to interpret.					
99	BLANK					



M-12 Coding Guide



Code	Response
Comple	te Response
10	40
Incorrec	t Response
70	30
79	Other incorrect.
Nonros	oonse
11011169	
<u>90</u>	Crossed out/erased, illegible, or impossible to interpret.





M-14 Coding Guide

Mirror

help you.

Μ 'N

M14. The picture shows a pencil that is lying on a shelf in front of a mirror. Draw a picture of the pencil as you would see it in the mirror. Use the patterns of lines on the shelf to

' D

В C G н

Κ *i* o Ρ

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		SN
		C XX
	a de atr	
		°O,
		
Code	C Response	
Correct	Response	
10	FGH; pencil point to the right.	
11	FGH; point not shown.	
12	FG or GH, (point to the right either shown or not shown) OR	
	any other in the row E,F,G, H as long as the point is not clearly	
	t Doopopoo	
	The sportse	
/0	in the row E F G H	
71	Lists all or some part of the row: ABCD.	
72	Lists all or some part of JKL; pencil point to the right may or	
	may not be shown.	
73	Lists all of some of the row MNOP, point to the right may or	
79	Other incorrect	
Norrasi		
inonres		
90	Crossed out/erased, illegible, or impossible to interpret.	
99	DLAINN	



Subject	Item Key	Content Category	Performance	Internationa Percent of Respondin	al Average Students g Correctly	International Difficulty
			Expectation	Upper Grade	Lower Grade	Index
Science	D	Environmental Issues and the Nature of Science	Investigating the Natural World	45%	40%	624



Subject	Item Key	Content Category	Performance	Percent of Respondin	Students g Correctly	International Difficulty
			Expectation	Upper Grade	Lower Grade	Index
Science	В	Life Science	Understanding Complex Information	42%	37%	648

N3. A cupful of water and a similar cupful of gasoline were placed on a table near a window on a hot sunny day. A few hours later it was observed that both the cups had less liquid in them but that there was less gasoline left than water. What does this experiment show?

All liquids evaporate.

Α.

B.

C.

- Gasoline gets hotter than water.
- Some liquids evaporate faster than others.
- Liquids will only evaporate in sunshine. D.

to to the the

Water gets hotter than gasoline E.

HUUL ENGR

Ministrant PXC

Subject	Item Key	Content Category	Performance	Internationa Percent of Respondin	al Average Students g Correctly	International Difficulty
			Expectation	Upper Grade	Lower Grade	Index
Science	С	Environmental Issues and the Nature of Science	Theorizing, Analyzing, and Solving Problems	62%	55%	526

ress



Subject	Item Key	Content Category	Performance	Percent of Respondin	Students g Correctly	International Difficulty
			Expectation	Upper Grade	Lower Grade	Index
Science	В	Life Science	Understanding Simple Information	50%	47%	584

N5. One of the principal causes of acid rain is

Β.

- waste acid from chemical factories being pumped into rivers A.
 - acid from chemical laboratories evaporating into the air

nistennine kritennine

- gases from burning coal and oil dissolving in water in the atmosphere
- gases from air conditioners and refrigerators escaping into the atmosphere

ress

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onnew express

Subject	Item Key	Content Category	Performance	Internationa Percent of Respondin	al Average Students g Correctly	International Difficulty
			Expectation	Upper Grade	Lower Grade	Index
Science	С	Environmental Issues and the Nature of Science	Understanding Simple Information	35%	31%	704



Subject	Item Key	Content Category	Performance	Percent of Respondin	Students g Correctly	International Difficulty
			Expectation	Upper Grade	Lower Grade	Index
Science	A	Life Science	Understanding Simple Information	67%	59%	503

N-6



Subject	Item Key	Content Category	Performance	Percent of Respondin	al Average Students g Correctly	International Difficulty
			Expectation	Upper Grade	Lower Grade	Index
Science	next page	Chemistry	Theorizing, Analyzing, and Solving Problems	89%	86%	291

N-7 Coding Guide



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produced from	TIMSS Population 2 Item Pool. Copyright © 1994 by IEA, The Hague	500
Code	Response K	only
Correct	Response	
10	Refers to the need for oxygen.	
	Examples: Fire does not get enough oxygen. The oxygen will be used up.	
11	Refers to the need for air. Example: Fire does not get enough air.	
12	Refers to the need for air, using non-scientific language. Examples: The fire will be "strangulated." The fire cannot breathe.	
19	Other correct.	
ncorrec	t Response	
70	Refers to its getting too hot	
71	States that the gas (smoke, vapor, carbon dioxide) is trapped inside the jar.	
72	Refers to the properties of the glass.	
76	Merely repeats the information in the stem.	
	Frample: The glass is placed over it	
79	Other incorrect:	
79	Other incorrect: Example: You put it on too fast and the wind makes it go out.	
79 Nonresi	Other incorrect: <i>Example: You put it on too fast and the wind makes it go out.</i> DONSE	
79 Nonresj 90	Other incorrect: <i>Example: You put it on too fast and the wind makes it go out.</i> ponse Crossed out/erased, illegible, or impossible to interpret.	



Subject	Item Key	Content Category	Performance	Internationa Percent of Respondin	al Average Students g Correctly	International Difficulty
			Expectation	Upper Grade	Lower Grade	Index
Science	D	Physics	Theorizing, Analyzing, and Solving Problems	72%	68%	459

N-8



Subject	Item Key	Content Category	Performance	Internationa Percent of Respondin	al Average Students g Correctly	International Difficulty
			Expectation	Upper Grade	Lower Grade	Index
Science	D	Chemistry	Using Tools, Routine Procedures, and Science Processes	52%	44%	605

N-9



Subject	Item Key	Content Category	Performance	Respondin	g Correctly	Difficulty
			Expectation	Upper Grade	Lower Grade	Index
Science	next page	Physics	Understanding Complex Information	52%	47%	569

N-10 Coding Guide

N10. A wat	ering can is almost filled with water as shown.		
The w	atering can is tipped so that the water just begins to drip through the spout.		
Draw	a line to show where the surface of the water in the can is now.		
Reproduced fror	n TIMSS Population 2 Item Pool. Copyright © 1994 by IEA, The Hague	ed	2
	ot be	2050	
	en navialexp	ion,	<
Code	Response Official Response	rom	£.
Code Correct	Response Response	rom	£,
Code Correct 10	Response Approximately* horizontal level of water within allowable range (see Figure 1).	ion i	£.
Code Correct 10	Response Approximately* horizontal level of water within allowable range (see Figure 1). Ct Response	ron	
Correct 10 Incorrec 70	Response Approximately* horizontal level of water within allowable range (see Figure 1). X Response Approximately* horizontal level of water. Higher or lower level of water than allowable range.	rom	
Code Correct 10 Incorrec 70 71	Response Approximately* horizontal level of water within allowable range (see Figure 1). C Response Approximately* horizontal level of water. Higher or lower level of water than allowable range. Water level is approximately* parallel to the bottom of the can (see Figure 2).	ron	
Code Correct 10 Incorrec 70 71 72	Response Approximately* horizontal level of water within allowable range (see Figure 1). CT Response Approximately* horizontal level of water. Higher or lower level of water than allowable range. Water level is approximately* parallel to the bottom of the can (see Figure 2). Water level clearly steeper than for code 71 (see Figure 3).	ron	
Code Correct 10 Incorrect 70 71 72 73	Response Approximately* horizontal level of water within allowable range (see Figure 1). X Response Approximately* horizontal level of water. Higher or lower level of water than allowable range. Water level is approximately* parallel to the bottom of the can (see Figure 2). Water level clearly steeper than for code 71 (see Figure 3). Water level is inclined in the opposite direction to that in codes 71 (see Figure 4).	rom	
Code Correct 10 Incorrec 70 71 72 73 79	Response Approximately* horizontal level of water within allowable range (see Figure 1). CT Response Approximately* horizontal level of water. Higher or lower level of water than allowable range. Water level is approximately* parallel to the bottom of the can (see Figure 2). Water level clearly steeper than for code 71 (see Figure 3). Water level is inclined in the opposite direction to that in codes 71 and 72 (see Figure 4). Other incorrect:	ion i	
Code Correct 10 Incorrect 70 71 72 73 79	Response Approximately* horizontal level of water within allowable range (see Figure 1). CT Response Approximately* horizontal level of water. Higher or lower level of water than allowable range. Water level is approximately* parallel to the bottom of the can (see Figure 2). Water level clearly steeper than for code 71 (see Figure 3). Water level is inclined in the opposite direction to that in codes 71 and 72 (see Figure 4). Other incorrect: Examples: Water in the spout only. Water only in the flower pot.	rom	
Correct 10 Incorrec 70 71 72 73 79 Nonres	Response Approximately* horizontal level of water within allowable range (see Figure 1). X Response Approximately* horizontal level of water. Higher or lower level of water than allowable range. Water level is approximately* parallel to the bottom of the can (see Figure 2). Water level clearly steeper than for code 71 (see Figure 3). Water level is inclined in the opposite direction to that in codes 71 and 72 (see Figure 4). Other incorrect: Examples: Water in the spout only. Water only in the flower pot. ponse	ron	
Code Correct 10 Incorrect 70 71 72 73 79 Nonres 90	Response Approximately* horizontal level of water within allowable range (see Figure 1). X Response Approximately* horizontal level of water. Higher or lower level of water than allowable range. Water level is approximately* parallel to the bottom of the can (see Figure 2). Water level clearly steeper than for code 71 (see Figure 3). Water level is inclined in the opposite direction to that in codes 71 and 72 (see Figure 4). Other incorrect: <i>Examples: Water in the spout only. Water only in the flower pot.</i> ponse Crossed out/erased, illegible, or impossible to interpret.	rom	

(Continued on next page.)

N-10 Coding Guide (Continued)




Information

page

O-10



O11. Which is a chemical change?

Β.

C

Element 1 is hammered into a thin sheet. A.

Element 2 is heated and turns into a liquid.

Element 3 turns a greenish color as it sits in air.

Element 4 is ground up into a fine, slippery powder. Je us ce

O-1

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Subject	ltem Key	Content Category	Performance	Percent of Respondin	al Average Students g Correctly	International Difficulty
			Expectation	Upper Grade	Lower Grade	Index
Science	С	Chemistry	Understanding Complex Information	38%	32%	670



O-12





and Solving Problems

page

O-14

O-14 Coding Guide

O14. The Sun is bigger than the Moon, but they appear to be about the same size when you look at them from the Earth. Why is this

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5 Stect	eo not be used not be used n
Code	Response
Correct	Response
10	Mentions that the sun is farther away than the moon.
	Comparative language is used.
	Example: The moon is closer to the Earth.
19	Other correct comparing apparent sizes.
Incorrec	t Response
70	Response includes some reference to the light.
	Examples: The sun shines on the moon.
	The moon shines only in the night.
71	States that the sun is closer than the moon.
72	Refers to distance, but response is general, not specified.
	Examples: We are so far away from the sun.
72	I ne aistance is so long.
74	Other incomplete or slightly erroneous responses
76	Merely repeats the information in the stem
	Frample: The sun is bigger than the moon
79	Other incorrect:
	Examples: Because you are in the same place.
	Because the sun is rotating.
Nonres	bonse
90	Crossed out/erased, illegible, or impossible to interpret.
99	BLANK





and Solving Problems

61%

54%

538

Science

Life Science

page

O-16

O-16 Coding Guide

O16. Write down the reason why we get thirsty on a hot day and have to drink a lot.

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Code Response 10 Refers to perspiration and its cooling effect and the need to replace lost water. 11 Refers to perspiration and only replacement of lost water. Example: Because when we are hot, out body opens the pores on our skin and we lose a lot of salt and liquid. 12 Refers to perspiration and only the scoling effect. 13 Refers to perspiration on dehydration only. Examples: We are sweating. Your body gives away much water. We are sweating and get drier. 19 Other acceptable explanation. Incorrect Response 70 Refers to body temperature (being too hot) but does not answer why we get thirsty. Example: You cool down by drinking something cold. 71 Refers only to drying of the body. Example: You get drier. 71 Refers to getting more energy by drinking more water. Example: You get drive. You get drive. 72 Refers to getting more energy by drinking more water. Example: You get exhausted. 76 Merely repeats the information in the stem. Example: You loose salt. Nonresponse 90 90 Crossed out/erased, illegible, or impossible to interpret. <t< th=""><th>5^{te^{ct}}</th><th>ee theurs</th><th>20 585</th></t<>	5 ^{te^{ct}}	ee theurs	20 585	
Correct Response 10 Refers to perspiration and only replacement of lost water. replace lost water. 11 Refers to perspiration and only replacement of lost water. Example: Because when we are hot, out body opens the pores on our skin and we lose a lot of salt and liquid. 12 Refers to perspiration and only its cooling effect. 13 Refers to perspiration or dehydration only. Examples: We are sweating. Your body gives away much water. We are sweating and get drier. 19 Other acceptable explanation. Incorrect Response 70 Refers to body temperature (being too hot) but does not answer why we get thirsty. Example: You cool down by drinking something cold. 71 Refers only to drying of the body. Examples: Your throat/mouth gets dry. You get drier. The heat dries everything. 72 Refers to getting more energy by drinking more water. Example: You get exhausted. 76 Merely repeats the information in the stem. Example: You loose salt. Nonresponse 90 90 Crossed out/erased, illegible, or impossible to interpret. 99 90 Crossed out/erased, illegible, or impossible to interpret.	Code	Response	5.	
10 Refers to perspiration and its cooling effect and the need to replace lost water. 11 Refers to perspiration and only replacement of lost water. Example: Because when we are hot, out body opens the pores on our skin and we lose a lot of salt and liquid. 12 Refers to perspiration and only its cooling effect. 13 Refers to perspiration or dehydration only. Example: We are sweating. Your body gives away much water. We are sweating and get drier. 0 19 Other acceptable explanation. Incorrect Response 70 Refers to body temperature (being too hot) but does not answer why we get thirsty. Example: You cool down by drinking something cold. 71 Refers only to drying of the body. Example: You throat/mouth gets dry. You get drier. The heat dries everything. 72 Refers to getting more energy by drinking more water. Example: You get exhausted. 76 Merely repeats the information in the stem. Example: You loose salt. Nonresponse 90 Crossed out/erased, illegible, or impossible to interpret. 99 BLANK	Correct	Response		
Instruction Image: Teplace lost water. Example: Because when we are hot, out body opens the pores on our skin and we lose a lot of salt and liquid. I2 Refers to perspiration and only its cooling effect. I3 Refers to perspiration or dehydration only. Example: We are sweating. Your body gives away much water. We are sweating and get drier. I9 Other acceptable explanation. Incorrect Response 70 Refers to body temperature (being too hot) but does not answer why we get thirsty. Example: You cool down by drinking something cold. 71 Refers only to drying of the body. Example: You get drier. The heat dries everything. The heat dries everything. 72 Refers to getting more energy by drinking more water. Example: You get exhausted. 76 Merely repeats the information in the stem. Example: Because it is hot. You need water. You need water. 79 Other incorrect: Example: You loose salt. Nonresponse 90 90 Crossed out/erased, illegible, or impossible to interpret. <th>10</th> <th>Refers to perspiration and its cooling effect and the need to</th> <th></th>	10	Refers to perspiration and its cooling effect and the need to		
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79 Other incorrect: Example: You loose salt. Nonresponse 90 Crossed out/erased, illegible, or impossible to interpret. 99 BLANK	=	You need water.		
Provide State Nonresponse 90 Crossed out/erased, illegible, or impossible to interpret. 99 BLANK	79	Other incorrect:		
90Crossed out/erased, illegible, or impossible to interpret.99BLANK	<u> </u>	Example. Tou loose sail.		
 90 Crossed out/erased, illegible, or impossible to interpret. 99 BLANK 	Nonres	ponse		
99 BLANK	90	Crossed out/erased, illegible, or impossible to interpret.		
	99	BLANK		



Information

57%

50%

556

Science

Life Science

page

O-17

O-17 Coding Guide

	O17. José cau	ght influenza. Write down one way he could have caught it.	
		× ·	
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			3
5			50
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)
			5 N
			C CV
			I
	Code	Résponse	
	Correct	Response	
	10	Refers explicitly to transmission of germs.	
	11	Refers implicitly to transmission of germs by sneezing/coughing	
		or close contact.	
		Shaking hands.	
		Eating together or from same utensils.	
	12	States only that he got it from someone (who had the flu).	
	19	<i>Example: He got it from someone else.</i>	
	Incorroo	t Bosponso	
		Refers to being too cold	
	10	<i>Examples: He got it from being out in the cold.</i>	
		He got it from getting wet [or freezing].	
	70	<i>He got it because he did not wear enough clothes.</i>	
	79	Utner incorrect.	
	Nonres		
	90	Crossed out/erased, illegible, or impossible to interpret.	
			l



Subject	Item Key	Content Category	Performance	Percent of Respondin	al Average Students g Correctly	International Difficulty
			Expectation	Upper Grade	Lower Grade	Index
Science	В	Physics	Using Tools, Routine Procedures, and Science Processes	83%	78%	358



Subject	Item Key	Content Category	Performance	Percent of Respondin	Students g Correctly	International Difficulty
			Expectation	Upper Grade	Lower Grade	Index
Science	next page	Physics	Theorizing, Analyzing, and Solving Problems	23%	18%	770



Jane and Mario were discussing what it might be like to live on other planets. P3. Their science teacher gave them data about the Earth and an imaginary planet, Athena. The table shows these data.

r	102	Earth	Athena
	A, O,	21% oxygen	10% oxygen
	Atmographia	0.03% carbon dioxide	80% carbon dioxide
	Conditions	78% nitrogen	5% nitrogen
	Conditions	ozone layer	no ozone layer
~	Distance from a Star	148,640,000 km	103,600,000 km
	Like the Sun		
	Rotation on Axis	1 day	200 days
	Revolution Around Sun	365 ¹ /4 days	200 days

Write down one important reason why it would be difficult for humans to live thouse Ares on Athena if it existed. Unithout exc

AISTENNING KAY COMME

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Subject	Item Key	Content Category	Performance	Internationa Percent of Respondin	al Average Students g Correctly	International Difficulty
			Expectation	Upper Grade	Lower Grade	Index
Science	next page	Earth Science	Theorizing, Analyzing, and Solving Problems	79%	75%	389

P-3 Coding Guide

Jane and Mario were discussing what it might be like to live on other planets. P3 Their science teacher gave them data about the Earth and an imaginary planet, Athena. The table shows these data. Earth Athena 21% oxygen 10% oxygen 0.03% carbon dioxide 80% carbon dioxide Atmospheric 78% nitrogen 5% nitrogen Conditions ozone layer no ozone layer Distance from a Star 148,640,000 km 103,600,000 km Like the Sun

200 days

	Rotation on Axis	1 day	200 days			
	Revolution Around Sun	365 ¹ / ₄ days	200 days			
	Write down one importat on Athena if it existed.	nt reason why it would be	difficult for humans to live		C	ed
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C	ode	<u>()</u>	Respons	;e		
Co	prrect Respor	nse		· · · · · · · · · · · · · · · · · · ·		
						4
	10 States the	at there would h	e too much carl	on dioxide		
	10 States the <i>Example</i>	at there would l	be too much carl	oon dioxide.		
	10States the Example11States the States the	at there would l : Too hot (great at there would l	be too much carl enhouse effect). be too little oxy	oon dioxide.		
	10 States the Example 11 States the Sta	at there would be <i>Too hot</i> (great at there would be bound rotation	be too much carl <u>enhouse effect</u>). be too little oxy 1, that is, the per	oon dioxide. gen to breathe. jods of revolution	on	
	10States the Example11States the Refers to around the	at there would be <u>Too hot (great</u> at there would bound rotation he planet's own	be too much carl enhouse effect). be too little oxy n, that is, the per axis and rotatio	oon dioxide. gen to breathe. iods of revolution n around its sun	on are the	
	10States the Example11States the States the around the same. He	at there would be <i>Too hot (great</i> at there would bound rotation he planet's own ence, one side	be too much carl <u>enhouse effect).</u> be too little oxy h, that is, the per- axis and rotation of the planet is a	oon dioxide. gen to breathe. iods of revolution n around its sun dways facing the	on are the e sun	
	10States the Example11States the States the around the same. He and there	at there would be <i>Too hot (great</i> at there would bound rotation he planet's own ence, one side fore is hot while	be too much carl <u>enhouse effect).</u> <u>be too little oxy</u> <u>n</u> , that is, the per <u>axis and rotation</u> of the planet is a le the other side	oon dioxide. gen to breathe. iods of revolution n around its sun ilways facing the is always dark an	on are the e sun ad cold.	
	10States the Example11States the States the around the same. He and there Example	at there would be <i>Too hot (green there would be there would be bound rotation to planet's own ence, one side of fore is hot while s: Too hot beck</i>	be too much carl enhouse effect). be too little oxy h, that is, the per axis and rotatio of the planet is a le the other side ause one side all	oon dioxide. gen to breathe. iods of revolution n around its sun dways facing the is always dark an ways faces the s	on are the e sun nd cold. <i>un</i> .	
	10States the Example11States the States the around the same. He and there Example Too cold	at there would be <i>Too hot (great there would be bound rotation to bound rotation the planet's own ence, one side of fore is hot whilk s: Too hot bect on the side that the side </i>	be too much carl enhouse effect). be too little oxy h, that is, the per axis and rotatio of the planet is a le the other side ause one side al this away from the	oon dioxide. gen to breathe. iods of revolution n around its sun dways facing the is always dark an ways faces the s he sun.	on are the e sun nd cold. <i>un</i> .	
	10States the Example11States the States the around the same. He and there Example Too cold13States the	at there would be <u>Too hot (great</u> at there would bound rotation he planet's own ence, one side offore is hot while s: Too hot beck on the side that at there is no office	be too much carl enhouse effect). be too little oxy h, that is, the per axis and rotatio of the planet is a le the other side ause one side al tt is away from t zone.	oon dioxide. gen to breathe. iods of revolution n around its sun dways facing the is always dark an ways faces the s he sun.	on are the e sun ad cold. <i>un</i> .	
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	10States the Example11States the states the around the same. He and there Example Too cold13States the Example14Any com	at there would be <i>Too hot (great</i> at there would bound rotation he planet's own ence, one side fore is hot whilk s: Too hot bect on the side that at there is no out <i>UV radiation</i> abination of about	be too much carl enhouse effect). be too little oxy h, that is, the per axis and rotatio of the planet is a le the other side ause one side al at is away from t zone. is dangerous. ove codes, 10-13	oon dioxide. gen to breathe. iods of revolution n around its sun lways facing the is always dark an ways faces the s he sun.	on are the e sun ad cold. <i>un</i> .	
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Inc	10States the Example11States the Example12Refers to around the same. He and there Example13States the Example14Any come19Other come70States the States the	at there would be <u>: Too hot (grea</u> at there would be bound rotation he planet's own ence, one side of fore is hot while s: Too hot beck on the side that at there is no of <u>: UV radiation</u> bination of about rect. DNSE at it is too close	be too much carl <u>enhouse effect).</u> be too little oxy h, that is, the per axis and rotation of the planet is a le the other side ause one side all tt is away from the zone. is dangerous. ove codes, 10-13 e to a star, without	oon dioxide. gen to breathe. iods of revolution n around its sun ilways facing the is always dark an ways faces the s he sun.	on are the e sun nd cold. <i>un</i> .	
Inc	10States the Example11States the Example12Refers to around the same. He and there Example13States the Example14Any com Other componence19Other componence Other componence70States the States the Other independence	at there would be <u>Too hot (great</u> at there would be bound rotation the planet's own ence, one side of fore is hot while s: Too hot bece on the side that at there is no of <u>CON the side that</u> there is no of <u>CON the side that</u> there is no of <u>CON the side that</u> at there is no of <u>CON the side that</u> there is no of <u>CON the side that</u> at there is no of <u>CON the side that</u> there is no of <u>CON the side that</u> there is no of <u>CON the side that</u> <u>CON the side that the side tha</u>	be too much carl <u>enhouse effect).</u> be too little oxy h, that is, the per- axis and rotation of the planet is a le the other side <u>ause one side all</u> <u>tt is away from the</u> zone. <u>is dangerous.</u> ove codes, 10-13 <u>e to a star, withe</u> usly incomplete.	oon dioxide. gen to breathe. iods of revolution n around its sun dways facing the is always dark an ways faces the s he sun.	on are the e sun nd cold. <i>un</i> .	
	10States the Example11States the Example12Refers to around the same. He and there Example13States the Example14Any come19Other come70States the Toole70States the Toole70States70States70States70States70States70States70States70States70States70States	at there would be <u>: Too hot (great</u> at there would be bound rotation he planet's own ence, one side of fore is hot while s: Too hot bect on the side that at there is no of <u>: UV radiation</u> bination of about rect. DNSE at it is too close correct or serior	be too much carl <u>enhouse effect).</u> be too little oxy h, that is, the per axis and rotatio of the planet is a le the other side ause one side all <u>tt is away from t</u> zone. <u>is dangerous.</u> ove codes, 10-13 e to a star, withous usly incomplete.	oon dioxide. gen to breathe. iods of revolution n around its sun dways facing the is always dark an ways faces the s he sun.	on are the e sun nd cold. <i>un</i> .	
	10States the Example11States the Example12Refers to around the same. He and there Example13States the Example14Any come19Other comecorrectResponse70States the Toold70States the Toold7	at there would be <u>Too hot (great</u> at there would be bound rotation the planet's own ence, one side of fore is hot while s: Too hot bece on the side that at there is no of <u>UV radiation</u> bination of about rect. DISE at it is too close correct or seriou out/erased, illes	be too much carl <u>enhouse effect).</u> be too little oxy h, that is, the per axis and rotatio of the planet is a le the other side <u>ause one side al</u> <u>tt is away from t</u> <u>is dangerous.</u> ove codes, 10-13 e to a star, withous usly incomplete.	oon dioxide. gen to breathe. iods of revolution n around its sun thways facing the is always dark an ways faces the s he sun. Dut further explan- tible to interpret.	on are the e sun nd cold. <i>un.</i>	

- P4. What happens when an animal hibernates?
 - There is no life in any of its parts. A.
 - It stops breathing. Β.

C

D.

E.

- Its temperature is higher than when it is active.
- It is absorbing energy for use when it is active. A CPC NING
- It is using less energy than when it is active.

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Subject	Item Key	Content Category	Performance	Internationa Percent of Respondin	al Average Students g Correctly	International Difficulty
			Expectation	Upper Grade	Lower Grade	Index
Science	Е	Life Science	Understanding Complex Information	56%	51%	559



Subject	Item Key	Content Category	Performance	Percent of Respondin	Students g Correctly	International Difficulty
			Expectation	Upper Grade	Lower Grade	Index
Science	next page	Physics	Theorizing, Analyzing, and Solving Problems	58%	52%	549

P-5 Coding Guide

P5. The wat	er in a tube is heated, as shown in the diagram. As the water is heated, the	
balloon	Increases in size. Explain why.	>
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2KOL	notbeur	20585
	This item man cial they are the connected to the connecte	es. ff
Code	Response	es. ff
Code Correct	Response	es-ff
Code Correct 10	Response Response Mentions explicitly that expansion is due to increased pressure of air/gas/water vapor when tube is heated.	es-ff
Code Correct 10	Response Mentions explicitly that expansion is due to increased pressure of air/gas/water vapor when tube is heated. States that the water evaporates.	es-ff
Code Correct 10 11 19	Response Response Mentions explicitly that expansion is due to increased pressure of air/gas/water vapor when tube is heated. States that the water evaporates. Other correct.	es. fr
Code Correct 10 11 19 Incorrect	Response Mentions explicitly that expansion is due to increased pressure of air/gas/water vapor when tube is heated. States that the water evaporates. Other correct. X Response	es-ff
Code Correct 10 11 19 Incorrect 70	Response Mentions explicitly that expansion is due to increased pressure of air/gas/water vapor when tube is heated. States that the water evaporates. Other correct. X Response Mentions that hot air[or gas] always rises.	esith
Code Correct 10 11 19 Incorrec 70 71	Response Mentions explicitly that expansion is due to increased pressure of air/gas/water vapor when tube is heated. States that the water evaporates. Other correct. X Response Mentions that hot air[or gas] always rises. Mentions that air particles [or molecules] expand when heated.	esith
Code Correct 10 11 19 Incorrect 70 71 76	Response Mentions explicitly that expansion is due to increased pressure of air/gas/water vapor when tube is heated. States that the water evaporates. Other correct. X Response Mentions that hot air[or gas] always rises. Mentions that air particles [or molecules] expand when heated. Merely repeats information in the stem.	esith
Code Correct 10 11 19 Incorrect 70 71 76	Response Mentions explicitly that expansion is due to increased pressure of air/gas/water vapor when tube is heated. States that the water evaporates. Other correct. Xt Response Mentions that hot air[or gas] always rises. Mentions that in particles [or molecules] expand when heated. Merely repeats information in the stem. Example: Because it is heated.	esith
Code Correct 10 11 19 Incorrect 70 71 76 79	Response Mentions explicitly that expansion is due to increased pressure of air/gas/water vapor when tube is heated. States that the water evaporates. Other correct. Xt Response Mentions that hot air[or gas] always rises. Mentions that hot air[or gas] always rises. Mentions that in particles [or molecules] expand when heated. Merely repeats information in the stem. Example: Because it is heated. Other incorrect.	esith
Code Correct 10 11 19 Incorrect 70 71 76 79 Nonres	Response Mentions explicitly that expansion is due to increased pressure of air/gas/water vapor when tube is heated. States that the water evaporates. Other correct. X Response Mentions that hot air[or gas] always rises. Mentions that hot air[or gas] always rises. Mentions that in particles [or molecules] expand when heated. Merely repeats information in the stem. Example: Because it is heated. Other incorrect. ponse	esith
Code Correct 10 11 19 Incorrect 70 71 76 79 Nonres 90	Response Mentions explicitly that expansion is due to increased pressure of air/gas/water vapor when tube is heated. States that the water evaporates. Other correct. X Response Mentions that hot air[or gas] always rises. Mentions that hot air[or gas] always rises. Mentions that air particles [or molecules] expand when heated. Merely repeats information in the stem. <i>Example: Because it is heated.</i> Other incorrect. ponse Crossed out/erased, illegible, or impossible to interpret.	esith



Subject	Item Key	Content Category	Performance	Percent of Students Responding Correctly		International Difficulty
			Expectation	Upper Grade	Lower Grade	Index
Science	next page	Life Science	Understanding Complex Information	40%	32%	636

P-6 Coding Guide

P6. W	hat digestive substance is found in the mouth? What does it do?	
	X . D.	
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	(° ° ° ° ° ° ° ° ° ° ° ° ° ° ° ° ° ° °	
Code	Response	on
Correc	e Response	on
Correct 20	Response Response Names saliva and explains that it makes the food moist or soft [mechanical process]. Example: Saliva. It helps us swallow.	on
Correc 20 21	B Response 2t Response Names saliva and explains that it makes the food moist or soft [mechanical process]. Example: Saliva. It helps us swallow. Names saliva and explains that it breaks down the starch or food. [Chemical process].	on
Code 20 21 22	Response X Response Names saliva and explains that it makes the food moist or soft [mechanical process]. Example: Saliva. It helps us swallow. Names saliva and explains that it breaks down the starch or food. [Chemical process]. Names enzymes and explains that they break down the starch or food. [Chemical process]. Names enzymes and explains that they break down the starch or food. [Chemical process].	on
Correc 20 21 22 29	e Response xt Response Names saliva and explains that it makes the food moist or soft [mechanical process]. Example: Saliva. It helps us swallow. Names saliva and explains that it breaks down the starch or food. [Chemical process]. Names enzymes and explains that they break down the starch or food. [Chemical process]. Other correct: Names a substance and provides a reasonable explanation.	
Correc 20 21 22 29 Partial	Response Xames saliva and explains that it makes the food moist or soft [mechanical process]. Example: Saliva. It helps us swallow. Names saliva and explains that it breaks down the starch or food. [Chemical process]. Names enzymes and explains that they break down the starch or food. [Chemical process]. Other correct: Names a substance and provides a reasonable explanation. Response	
Correc 20 21 22 29 Partial 10	e Response xt Response Names saliva and explains that it makes the food moist or soft [mechanical process]. Example: Saliva. It helps us swallow. Names saliva and explains that it breaks down the starch or food. [Chemical process]. Names enzymes and explains that they break down the starch or food. [Chemical process]. Other correct: Names a substance and provides a reasonable explanation. Response Names saliva but with no description or with an incorrect description of what it does. Example: Saliva. Contains acid which helps digesting the food.	
Code 20 21 22 29 Partial 10 11	e Response xt Response Names saliva and explains that it makes the food moist or soft [mechanical process]. Example: Saliva. It helps us swallow. Names saliva and explains that it breaks down the starch or food. [Chemical process]. Names enzymes and explains that they break down the starch or food. [Chemical process]. Other correct: Names a substance and provides a reasonable explanation. Response Names saliva but with no description or with an incorrect description of what it does. Example: Saliva. Contains acid which helps digesting the food. Names enzymes but with no description or with an incorrect description of what they do, such as It digests starch.	
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Code 20 21 22 29 Partial 10 11 19 Incorrect 70	e Response tt Response Names saliva and explains that it makes the food moist or soft [mechanical process]. Example: Saliva. It helps us swallow. Names saliva and explains that it breaks down the starch or food. [Chemical process]. Names enzymes and explains that they break down the starch or food. [Chemical process]. Other correct: Names a substance and provides a reasonable explanation. Response Names saliva but with no description or with an incorrect description of what it does. Example: Saliva. Contains acid which helps digesting the food. Names enzymes but with no description or with an incorrect description of what it does. Example: Saliva. Contains acid which helps digesting the food. Names enzymes but with no description or with an incorrect description of what they do, such as It digests starch. Other partially correct. Set Response Acid. With or without description.	
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Code 20 21 22 29 Partial 10 11 19 Incorrect 70 71 79	e Response ct Response Names saliva and explains that it makes the food moist or soft [mechanical process]. Example: Saliva. It helps us swallow. Names saliva and explains that it breaks down the starch or food. [Chemical process]. Names enzymes and explains that they break down the starch or food. [Chemical process]. Other correct: Names a substance and provides a reasonable explanation. Response Names saliva but with no description or with an incorrect description of what it does. Example: Saliva. Contains acid which helps digesting the food. Names enzymes but with no description or with an incorrect description of what they do, such as It digests starch. Other partially correct. ect Response Acid. With or without description. Teeth, tongue, etc. With or without description. Other incorrect.	
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P7. Whenever scientists carefully measure any quantity many times, they expect that

A. All of the measurements will be exactly the same

nistenninne korconne

Β.

C.

D.

only two of the measurements will be exactly the same

all but one of the measurements will be exactly the same

most of the measurements will be close but not exactly the same

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man claiphess frank

Subject	ltem Key	Content Category	Performance	International Average Percent of Students Responding Correctly		International Difficulty
			Expectation	Upper Grade	Lower Grade	Index
Science	D	Environmental Issues and the Nature of Science	Understanding Simple Information	53%	49%	570



Understanding Complex

Information

Science

А

Earth Science

Upper Grade

44%

Lower Grade

39%

627



and Solving Problems

page

Q-12 Coding Guide

O12. Jim and	Sandy each make a flashlight from identical batteries and bulbs. Sandy's	
flashligh	t contains a reflector, while Jim's does not.	
Jīm's	Itashlight Sandy's flashlight	
Which fl	ashlight shines more light on a wall 5 meters away?	
(check or	ne)	
	lim's	
	Sandy's	$\mathbf{\lambda}$
Explain	your answer.	<u>,</u> ,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,
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	this tenner ciale the	ont
	this item mercial expression	ont
Code	Response	omit
Code Correct	Response	ont
Code Correct 10	Response Sandy's. The reflector reflects all the light towards the wall.	ont
Code Correct 10 11	Response Sandy's. The reflector reflects all the light towards the wall. Sandy's. In Jim's flashlight the light shines in all directions.	ont
Code Correct 10 11 12	Response Sandy's. The reflector reflects all the light towards the wall. Sandy's. In Jim's flashlight the light shines in all directions. Any combination of codes 10, 11.	ont
Code Correct 10 11 12 19	Response Sandy's. The reflector reflects all the light towards the wall. Sandy's. In Jim's flashlight the light shines in all directions. Any combination of codes 10, 11. Sandy's. Other correct explanations.	ont
Correct 10 11 12 19 Incorrect	Response Sandy's. The reflector reflects all the light towards the wall. Sandy's. In Jim's flashlight the light shines in all directions. Any combination of codes 10, 11. Sandy's. Other correct explanations. t Response	ont
Correct 10 11 12 19 Incorrect 70	Response Sandy's. The reflector reflects all the light towards the wall. Sandy's. In Jim's flashlight the light shines in all directions. Any combination of codes 10, 11. Sandy's. Other correct explanations. Any combination of codes 10, 11. Sandy's. Other correct explanations.	ont
Code Correct 10 11 12 19 Incorrect 70	Response Sandy's. The reflector reflects all the light towards the wall. Sandy's. The reflector reflects all the light shines in all directions. Any combination of codes 10, 11. Sandy's. Other correct explanations. t Response Jim's. Examples: There is no reflector. It is not covered	ont
Code Correct 10 11 12 19 Incorrect 70	Response Sandy's. The reflector reflects all the light towards the wall. Sandy's. The reflector reflects all the light towards the wall. Sandy's. In Jim's flashlight the light shines in all directions. Any combination of codes 10, 11. Sandy's. Other correct explanations. t Response Jim's. Examples: There is no reflector. It is not covered. Iim's but with an explanation that belongs to Sandy's	ont
Code Correct 10 11 12 19 Incorrect 70 71 72	Response Sandy's. The reflector reflects all the light towards the wall. Sandy's. The reflector reflects all the light towards the wall. Sandy's. In Jim's flashlight the light shines in all directions. Any combination of codes 10, 11. Sandy's. Other correct explanations. t Response Jim's. Examples: There is no reflector. It is not covered. Jim's but with an explanation that belongs to Sandy's. Iim's with another explanation	ont
Code Correct 10 11 12 19 Incorrect 70 71 72 73	Response Sandy's. The reflector reflects all the light towards the wall. Sandy's. The reflector reflects all the light towards the wall. Sandy's. In Jim's flashlight the light shines in all directions. Any combination of codes 10, 11. Sandy's. Other correct explanations. t Response Jim's. Examples: There is no reflector. It is not covered. Jim's but with an explanation that belongs to Sandy's. Jim's or Sandy's, with no explanation.	ont
Code Correct 10 11 12 19 Incorrect 70 71 72 73 76	Response Sandy's. The reflector reflects all the light towards the wall. Sandy's. In Jim's flashlight the light shines in all directions. Any combination of codes 10, 11. Sandy's. Other correct explanations. X Response Jim's. Examples: There is no reflector. It is not covered. Jim's or Sandy's, with no explanation. Jim's or Sandy's, with no explanation. Merely repeats information in stem.	ont
Code Correct 10 11 12 19 Incorrect 70 71 72 73 76	Response Sandy's. The reflector reflects all the light towards the wall. Sandy's. In Jim's flashlight the light shines in all directions. Any combination of codes 10, 11. Sandy's. Other correct explanations. t Response Jim's. Examples: There is no reflector. It is not covered. Jim's but with an explanation that belongs to Sandy's. Jim's or Sandy's, with no explanation. Merely repeats information in stem. Example: Sandy's, because of the reflector.	ont
Code Correct 10 11 12 19 Incorrect 70 71 72 73 76 79	Response Sandy's. The reflector reflects all the light towards the wall. Sandy's. The reflector reflects all the light towards the wall. Sandy's. In Jim's flashlight the light shines in all directions. Any combination of codes 10, 11. Sandy's. Other correct explanations. X Response Jim's. <i>Examples: There is no reflector.</i> It is not covered. Jim's but with an explanation that belongs to Sandy's. Jim's or Sandy's, with no explanation. Merely repeats information in stem. <i>Example: Sandy's, because of the reflector.</i> Other incorrect.	ont
Code Correct 10 11 12 19 Incorrect 70 71 72 73 76 79 Nonres	Response Sandy's. The reflector reflects all the light towards the wall. Sandy's. The reflector reflects all the light towards the wall. Sandy's. In Jim's flashlight the light shines in all directions. Any combination of codes 10, 11. Sandy's. Other correct explanations. Xt Response Jim's. Examples: There is no reflector. It is not covered. Jim's but with an explanation that belongs to Sandy's. Jim's or Sandy's, with no explanation. Merely repeats information in stem. Example: Sandy's, because of the reflector. Other incorrect. Donse	ont
Code Correct 10 11 12 19 Incorrec 70 71 72 73 76 79 Nonres 90	Response Sandy's. The reflector reflects all the light towards the wall. Sandy's. The reflector reflects all the light shines in all directions. Any combination of codes 10, 11. Sandy's. Other correct explanations. X Response Jim's. Examples: There is no reflector. It is not covered. Jim's but with an explanation that belongs to Sandy's. Jim's vor Sandy's, with no explanation. Merely repeats information in stem. Example: Sandy's, because of the reflector. Other incorrect. Consee Crossed out/erased, illegible, or impossible to interpret.	ont





Understanding Complex

Information

46%

37%

613

Science

Е

Chemistry



Understanding Simple

Information

31%

26%

693

Science

А

Chemistry



Understanding Simple

Information

27%

23%

740

Science

D

Earth Science



Subject	Item Key	Content Category	Performance Expectation	Percent of Students Responding Correctly		International Difficulty
				Upper Grade	Lower Grade	Index
Science	next page	Life Science	Understanding Complex Information	64%	59%	493

Q-17 Coding Guide





Theorizing, Analyzing,

and Solving Problems

29%

24%

665

next

page

Physics

Science

Q-18 Coding Guide

ALOK C

Q18. A glass of water with ice cubes in it has a mass of 300 grams. What will the mass be immediately after the ice has melted? Explain your answer.

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Note: For this question do not distinguish if the student substitutes kg for g: that is, accept 300 kg as the same as 300 g.

peuseus , moses

, v,

Code	Response
Correct	Response
20	300 g with a good explanation.
	Examples: 300 g. The ice changes into the same amount of water.
	The same. The ice only melts.
	The same weight. Nothing disappears.
Partial F	Response
10	300 g. Explanation is inadequate.
11	300 g. No explanation.
Incorrec	tResponse
70	More than 300 grams with explanation.
	Examples: More. Water has higher density.
	More. Water is heavier than ice
71	More than 300 g. No explanation.
72	Less than 300 g. With explanation.
	Examples: Less. Ice is heavier than water.
	Less. There will be water only.
73	Less than 300 g. No explanation.
79	Other incorrect.
Nonres	oonse
90	Crossed out/erased, illegible, or impossible to interpret.
99	BLANK



Understanding Simple

Information

72%

66%

468

Science

А

Physics

R-1

When white light shines on Peter's shirt, the shirt looks blue. Why does the R2. shirt look blue?

It absorbs all the white light and turns most of it into blue light.

It reflects the blue part of the light and absorbs most of the rest.

It absorbs only the blue part of the light.

AISTENNING KAY COMME

It gives off its own blue light.

А.

Β.

C.

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mercial purposes

Subject	Item Key	Content Category	Performance	International Average Percent of Students Responding Correctly		International Difficulty
			Expectation	Upper Grade	Lower Grade	Index
Science	В	Physics	Understanding Simple Information	39%	35%	653

R-2


Information

13%

9%

807

Science

Life Science

page

R-3

R-3 Coding Guide

R3. What co area? G	ould be the unwanted consequences of introducing a new species to a certain tive an example.	
	× ·	
Reproduced from	n TIMSS Population 2 Item Pool. Copyright © 1994 by IEA, The Hague	2025
	21.21	2.1
Codo	ten nercie exer	omli
Code	Response	onli
Code Correct	Response Response	onli
Code Correct 20	Response States that the natural (ecological) balance will be upset. A realistic example of a species is given.	onli
Code Correct 20 21	Response States that the natural (ecological) balance will be upset. A realistic example of a species is given. States that the new species may take over and gives examples. Example: It may kill everything. If you put an alligator in a fish and duck pond, it will eat the fish and ducks.	on
Code Correct 20 21 29	Response States that the natural (ecological) balance will be upset. A realistic example of a species is given. States that the new species may take over and gives examples. Example: It may kill everything. If you put an alligator in a fish and duck pond, it will eat the fish and ducks. Other correct responses with examples.	onli
Correct 20 21 29 Partial I	Response States that the natural (ecological) balance will be upset. A realistic example of a species is given. States that the new species may take over and gives examples. Example: It may kill everything. If you put an alligator in a fish and duck pond, it will eat the fish and ducks. Other correct responses with examples. Response	onli
Code Correct 20 21 29 Partial I 10	Response States that the natural (ecological) balance will be upset. A realistic example of a species is given. States that the new species may take over and gives examples. Example: It may kill everything. If you put an alligator in a fish and duck pond, it will eat the fish and ducks. Other correct responses with examples. Response Adequate explanation (as in codes 20, 21), but no concrete and realistic example is given. Example: There will be too many.	on
Correct 20 21 29 Partial I 10	Response States that the natural (ecological) balance will be upset. A realistic example of a species is given. States that the new species may take over and gives examples. Example: It may kill everything. If you put an alligator in a fish and duck pond, it will eat the fish and ducks. Other correct responses with examples. Response Adequate explanation (as in codes 20, 21), but no concrete and realistic example is given. Example: There will be too many. Only the realistic example is given, but no explanation.	om
Correct 20 21 29 Partial I 10 11 12	Response States that the natural (ecological) balance will be upset. A realistic example of a species is given. States that the new species may take over and gives examples. Example: It may kill everything. If you put an alligator in a fish and duck pond, it will eat the fish and ducks. Other correct responses with examples. Response Adequate explanation (as in codes 20, 21), but no concrete and realistic example is given. Example: There will be too many. Only the realistic example is given, but no explanation. States the new species cannot live here.	onli
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Code Correct 20 21 29 Partial I 10 11 12 19 Incorrect 70 79	Response States that the natural (ecological) balance will be upset. A realistic example of a species is given. States that the natural (ecological) balance will be upset. A realistic example of a species is given. States that the new species may take over and gives examples. Example: It may kill everything. If you put an alligator in a fish and duck pond, it will eat the fish and ducks. Other correct responses with examples. Response Adequate explanation (as in codes 20, 21), but no concrete and realistic example is given. Example: There will be too many. Only the realistic example is given, but no explanation. States the new species cannot live here. Other partially correct. Ct Response Only an unrealistic example is given. Example: Introducing polar bears into the Sahara. Other incorrect.	om
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Information

page

R-4

R-4 Coding Guide

verected votected

R4. Write down one reason why the ozone layer is important for all living things on Earth.

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Code	Response	,0
Correct	Response	
10	Refers to protection against the UV radiation from the sun.	
11	Refers to protection against dangerous or too strong radiation	
	from the sun but does not mention UV.	
	<i>Example:</i> Because it keeps the sun's rays from being too strong.	
12	Mentions that the ozone layer protects humans so we do not get	
10	sunburned/skin cancer. NOTE: If UV is mentioned, code 10.	
17	t Deenenee	
Incorrec	t Response	
70	Confuses the effect of the ozone layer with the greenhouse	
=1	Example: It keeps the heat in.	
71	Confuses protection against heat.	
70	Example: Everything will melt without it.	
12	Kerers only vaguely to protection.	
	Examples: All living things will ale without the ozone layer.	
72	It protects the Edith/us.	
13	Frample: It is needed for respiration	
74	Sans the ozona layar as a harriar for the streamhara	
/4	Example: It keeps the air around the earth	
76	Merely repeats information in the stem	
70	Other incorrect	
Nonres		
Nonres		
<u> </u>	Crossed out/erased, illegible, or impossible to interpret.	



Theorizing, Analyzing,

and Solving Problems

50%

42%

589

next

page

Chemistry

Science

R-5 Coding Guide

R5. Carbon dioxide	lioxide is the active material in some fire extinguishers. How does carbon	
uloxide		
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	is tenner cia ex	Romit
	this ten mercia et	Riom in
Code	Response dia	
Code Correct	Response	
Code Correct 10	Response Mentions that carbon dioxide keeps oxygen away; response	
Code Correct 10	Response Mentions that carbon dioxide keeps oxygen away; response includes explicit reference to oxygen. Framework for desard act oxygen.	
Code Correct 10	Response Response Response Response Sector of the sector of	
Code Correct 10	Response Response Mentions that carbon dioxide keeps oxygen away; response includes explicit reference to oxygen. Examples: Fire doesn't get oxygen. Carbon dioxide is heavier than oxygen and will cover the burning materials.	
Code Correct 10	Response Response Mentions that carbon dioxide keeps oxygen away; response includes explicit reference to oxygen. Examples: Fire doesn't get oxygen. Carbon dioxide is heavier than oxygen and will cover the burning materials. Mentions that carbon dioxide keeps "air" away	
Code Correct 10	Response Response Mentions that carbon dioxide keeps oxygen away; response includes explicit reference to oxygen. Examples: Fire doesn't get oxygen. Carbon dioxide is heavier than oxygen and will cover the burning materials. Mentions that carbon dioxide keeps "air" away Examples: It strangulates the flame	
Code Correct 10	Response Response Response Mentions that carbon dioxide keeps oxygen away; response includes explicit reference to oxygen. Examples: Fire doesn't get oxygen. Carbon dioxide is heavier than oxygen and will cover the burning materials. Mentions that carbon dioxide keeps "air" away Examples: It strangulates the flame The fire cannot "breathe."	
Code Correct 10	Response Response Mentions that carbon dioxide keeps oxygen away; response includes explicit reference to oxygen. Examples: Fire doesn't get oxygen. Carbon dioxide is heavier than oxygen and will cover the burning materials. Mentions that carbon dioxide keeps "air" away Examples: It strangulates the flame The fire cannot "breathe." Other correct.	
Code Correct 10 11 19 Incorrec	Response Response Mentions that carbon dioxide keeps oxygen away; response includes explicit reference to oxygen. Examples: Fire doesn't get oxygen. Carbon dioxide is heavier than oxygen and will cover the burning materials. Mentions that carbon dioxide keeps "air" away Examples: I strangulates the flame The fire cannot "breathe." Other correct. X Response	
Code Correct 10 11 19 Incorrec 70	Response Response Mentions that carbon dioxide keeps oxygen away; response includes explicit reference to oxygen. Examples: Fire doesn't get oxygen. Carbon dioxide is heavier than oxygen and will cover the burning materials. Mentions that carbon dioxide keeps "air" away Examples: It strangulates the flame The fire cannot "breathe." Other correct. Xt Response	
Code Correct 10 11 19 Incorrec 70 71	Response Response Mentions that carbon dioxide keeps oxygen away; response includes explicit reference to oxygen. Examples: Fire doesn't get oxygen. Carbon dioxide is heavier than oxygen and will cover the burning materials. Mentions that carbon dioxide keeps "air" away Examples: It strangulates the flame The fire cannot "breathe." Other correct. X Response Mentions that carbon dioxide cools down the fire. Refers to a material in carbon dioxide.	
Code Correct 10 11 19 Incorrect 70 71 79	Response Response Mentions that carbon dioxide keeps oxygen away; response includes explicit reference to oxygen. Examples: Fire doesn't get oxygen. Carbon dioxide is heavier than oxygen and will cover the burning materials. Mentions that carbon dioxide keeps "air" away Examples: It strangulates the flame The fire cannot "breathe." Other correct. X Response Mentions that carbon dioxide cools down the fire. Refers to a material in carbon dioxide. Other incorrect.	
Code Correct 10 11 19 Incorrect 70 71 79	Response Response Mentions that carbon dioxide keeps oxygen away; response includes explicit reference to oxygen. Examples: Fire doesn't get oxygen. Carbon dioxide is heavier than oxygen and will cover the burning materials. Mentions that carbon dioxide keeps "air" away Examples: It strangulates the flame The fire cannot "breathe." Other correct. Xt Response Mentions that carbon dioxide cools down the fire. Refers to a material in carbon dioxide. Other incorrect.	
Code Correct 10 11 19 Incorrec 70 71 79 Nonres	Response Response Mentions that carbon dioxide keeps oxygen away; response includes explicit reference to oxygen. Examples: Fire doesn't get oxygen. Carbon dioxide is heavier than oxygen and will cover the burning materials. Mentions that carbon dioxide keeps "air" away Examples: It strangulates the flame The fire cannot "breathe." Other correct. X Response Mentions that carbon dioxide cools down the fire. Refers to a material in carbon dioxide. Other incorrect. ponse Crassed out/accord, illegible, extimpose ible to interpret	
Correct 10 11 19 Incorrec 70 71 79 Nonres 90 00	Response Response Mentions that carbon dioxide keeps oxygen away; response includes explicit reference to oxygen. Examples: Fire doesn't get oxygen. Carbon dioxide is heavier than oxygen and will cover the burning materials. Mentions that carbon dioxide keeps "air" away Examples: It strangulates the flame The fire cannot "breathe." Other correct. X Response Mentions that carbon dioxide cools down the fire. Refers to a material in carbon dioxide. Other incorrect. ponse Crossed out/erased, illegible, or impossible to interpret.	



	Subject	Item Key	Content Category	Performance	Responding Correctly		Difficult
				Expectation	Upper Grade	Lower Grade	Index
•	Science	next page	Earth Science	Theorizing, Analyzing, and Solving Problems	79%	76%	383

W-1a Coding Guide



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A: Codes for reason plain is a good place for farming.

Code	Response
Correct	Response
10	Mentions that the soil is fertile (good), abundant.
11	Mentions that there is a river (for irrigation, water for animals).
12	Mentions that there is plenty of space or flat areas for farm land.
19	Other correct:
	Example: The goats can find grass in the mountains.
Incorrec	t Response
70	Does not address the issue of farming.
	<i>Examples:</i> It is silent, a peaceful place to live.
	<i>Examples: It is silent, a peaceful place to live.</i> <i>You can swim in the river.</i>
76	Examples: It is silent, a peaceful place to live. You can swim in the river. Merely repeats information in stem.
76 79	Examples: It is silent, a peaceful place to live. You can swim in the river. Merely repeats information in stem. Other incorrect.
76 79 Nonres	Examples: It is silent, a peaceful place to live. You can swim in the river. Merely repeats information in stem. Other incorrect.
76 79 Nonres 90	Examples: It is silent, a peaceful place to live. You can swim in the river. Merely repeats information in stem. Other incorrect. OONSE Crossed out/erased, illegible, or impossible to interpret.

مى مەر



W-1b Coding Guide



Ь.	Write down one reason why this plain is NOT a good place for farming.	e
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B: Codes	s for reason plain is not a good place for farming.	SUF
Code	Response	
Correct	Response	
10	Mentions the possibility of flooding, or that the soil will be too	
10	wet	
11	Mentions the possibility of wind or water erosion.	
19	Other correct:	
	Examples: They might not get a lot of sunlight.	
	The farmer would have to climb the hills to sell or tradehis meat.	
	vegetables, or fruit.	
	It might be in the rain shadow of one of the mountains or hills.	
Incorrec	ct Response	
70	Mentions that it is an undesirable place to live:	
70	horing/lonesome/ugly	
	Example: Too far from the city.	
71	Does not address the issue of farming.	
	<i>Example:</i> The river is dangerous [for children].	
72	Refers to problems due to surrounding mountains.	
	<i>Examples:</i> Avalanches (snow or rocks) from the mountains.	
	Goats get lost in the mountains.	
73	Refers to sediment, soil, being rocky and negative.	
76	Merely repeats information in stem.	
79	Other incorrect.	
Nonres	sponse	
90	Crossed out/erased, illegible, or impossible to interpret.	
00	BLANK	



Subject	ltem Key	Content Category	Performance	International Average Percent of Students Responding Correctly		International Difficulty
			Expectation	Upper Grade	Lower Grade	Index
Science	next page	Earth Science	Theorizing, Analyzing, and Solving Problems	32%	27%	659

W-2

W-2 Coding Guide





Subject	Item Key	em Key Content Category	Performance	Percent of Students Responding Correctly		International Difficulty
			Expectation	Upper Grade	Lower Grade	Index
Science	next page	Life Science	Investigating the Natural World	14%	8%	797

X-1

X-1 Coding Guide

X1. Suppose you want to investigate how the human heart rate changes with changes in activity. What materials would you use and what procedures would you follow?

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X	
X	
3	
	NY SIN
Code	Response
Correct	Response
20	Describes a procedure in which:
	i. Somebody (or self) measures pulse at rest using a timer or
	watch.
	ii. Then student does an exercise (physical activity).
	111. Pulse is remeasured during or after the exercise.
	activities
Partial I	Response
10	As in code 20, but there is no mention of two or more
10	measurements.
11	As in code 20, but there is no reference to any exercise.
12	As in code 20, but there is no reference to a timer, watch, etc.
19	Other partially correct.
Incorrec	ct Response
70	Describes procedure but omits mention of two or more of the
	criteria stated in code 20.
79	Other incorrect.
Nonres	ponse
90	Crossed out/erased, illegible, or impossible to interpret.
99	BLANK



	Subject	Item Key	Content Category	Performance Expectation	Internationa Percent of Respondin Upper Grade	al Average Students g Correctly Lower Grade	International Difficulty Index
•	Science	next page	Life Science	Theorizing, Analyzing, and Solving Problems	64%	58%	474

V Do Codina C . .

X2. In the p	picture of an aquarium, six items are labeled.	
RY	Thetmometer Light Castle Plant Plant Rock Snail	>
aquariu	m.	
(b) the	light	ses
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		6
A: Codes	for Plant	0
A: Codes	for Plant Response	0
A: Codes Code Correct	for Plant Response	0
A: Codes Code Correct 10	for Plant Response Plants produce oxygen (by photosynthesis).	0
A: Codes Code Correct 10 11	for Plant Response Plants produce oxygen (by photosynthesis). Plants clean the water.	0
A: Codes Code Correct 10 11 12	for Plant Response Plants produce oxygen (by photosynthesis). Plants clean the water. Plants provide food for the fish.	0
A: Codes Code Correct 10 11 12 13	for Plant Response Plants produce oxygen (by photosynthesis). Plants clean the water. Plants provide food for the fish. Plants provide a place to hide/shelter, or to hide eggs.	0
A: Codes Code Correct 10 11 12 13 19	for Plant Response Plants produce oxygen (by photosynthesis). Plants clean the water. Plants provide food for the fish. Plants provide a place to hide/shelter, or to hide eggs. Other correct.	0
A: Codes Correct 10 11 12 13 19 Incorrect	for Plant Response Plants produce oxygen (by photosynthesis). Plants clean the water. Plants provide food for the fish. Plants provide a place to hide/shelter, or to hide eggs. Other correct. tt Response	0
A: Codes Correct 10 11 12 13 19 Incorrec 70	for Plant Response Plants produce oxygen (by photosynthesis). Plants provide food for the fish. Plants provide a place to hide/shelter, or to hide eggs. Other correct. t Response Plants improve the natural surroundings.	0
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tem Key	Content Category	Performance	Responding Correctly		
		Expectation	Upper Grade	Lower Grade	
next page	Life Science	Theorizing, Analyzing, and Solving Problems	33%	26%	

Science

116

Index

685

X-2b Coding Guide



V1	Electrical energy is used to never a lower	
	Electrical energy is used to power a lamp.	
	Is the amount of light energy produced more than	n, less than, or the same as
	the amount of electrical energy used?	
	The amount of light energy produced is	
	more than	
0		
	less than	(check one)
	the same as	50 6
XO.	the amount of electrical energy used.	0. 0
R		V° 0'
	Give a reason to support your answer.	Nº SN.
		X , C , K
		etion
	.×C/. (
		i.V
	A A A A	2
	eli	
	X	
		Alter Deal Consider 10041 The The V
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1		international Average

Subject	Subject Item Key Content Category Performance		Percent of Students Responding Correctly		International Difficulty	
			Expectation	Upper Grade	Lower Grade	Index
Science	next page	Physics	Theorizing, Analyzing, and Solving Problems	8%	4%	963

Y-1

Y-1 Coding Guide

Y1.	Electrica	l energy is used to power a lamp.	
	Is the arr	nount of light energy produced more than, less than, or the same as	
	the amou	int of electrical energy used?	
	The amo	unt of light energy produced is	
	m	ore than	
	le	ss than	
		CHECK ONE)	
	th	e amount of electrical energy used.	
Ψ	· .		
	Give a re	ason to support your answer.	
	2		
			5
		not pe	ipose i
		is item mercia et	Pronte
Co	ode	Response	
Co	ode rrect	Response	
Co	ode rrect	Response Less. Mentions that (much) energy is transformed to heat. Loss Mentions that anorgy is precided to warm up the lower	
Co	ode rrect 10 11 12	Response Less. Mentions that (much) energy is transformed to heat. Less. Mentions that energy is needed to warm up the lamp. Less. Mentions that energy (heat) is lost to the surroundings	
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	o de rrect 10 11 12 19 orrec 70	Response Less. Mentions that (much) energy is transformed to heat. Less. Mentions that energy is needed to warm up the lamp. Less. Mentions that energy (heat) is lost to the surroundings. Less. Other correct. t Response The same. With erroneous explanation. Examples: Energy is always preserved	
Co Inc	o de rrect 10 11 12 19 orrec 70	Response Less. Mentions that (much) energy is transformed to heat. Less. Mentions that energy is needed to warm up the lamp. Less. Mentions that energy (heat) is lost to the surroundings. Less. Other correct. t Response The same. With erroneous explanation. Examples: Energy is always preserved. When the sun is out you don't need electrical energy.	
Co Inc	ode <u>rrect</u> <u>10</u> <u>11</u> <u>12</u> <u>19</u> <u>orrec</u> <u>70</u> <u>71</u>	Response Less. Mentions that (much) energy is transformed to heat. Less. Mentions that energy is needed to warm up the lamp. Less. Mentions that energy (heat) is lost to the surroundings. Less. Other correct. t Response The same. With erroneous explanation. Examples: Energy is always preserved. When the sun is out you don't need electrical energy. The same. No explanation is given.	
	ode rrect 10 11 12 19 orrec 70 71 72	Response Less. Mentions that (much) energy is transformed to heat. Less. Mentions that energy is needed to warm up the lamp. Less. Mentions that energy is needed to warm up the lamp. Less. Mentions that energy (heat) is lost to the surroundings. Less. Other correct. t Response The same. With erroneous explanation. Examples: Energy is always preserved. When the sun is out you don't need electrical energy. The same. No explanation is given. More. With or without explanation.	
	D de <u>rrect</u> <u>10</u> <u>11</u> <u>12</u> <u>19</u> <u>0</u> <u>70</u> <u>71</u> <u>72</u> <u>73</u>	Response Less. Mentions that (much) energy is transformed to heat. Less. Mentions that energy is needed to warm up the lamp. Less. Mentions that energy is needed to warm up the lamp. Less. Mentions that energy (heat) is lost to the surroundings. Less. Other correct. t Response The same. With erroneous explanation. Examples: Energy is always preserved. When the sun is out you don't need electrical energy. The same. No explanation is given. More. With or without explanation. Less. No explanation.	
	D de rrect 10 11 12 19 Orrec 70 71 72 73 74	Response Less. Mentions that (much) energy is transformed to heat. Less. Mentions that energy is needed to warm up the lamp. Less. Mentions that energy is needed to warm up the lamp. Less. Mentions that energy (heat) is lost to the surroundings. Less. Other correct. t Response The same. With erroneous explanation. Examples: Energy is always preserved. When the sun is out you don't need electrical energy. The same. No explanation is given. More. With or without explanation. Less. No explanation. Less. No explanation. Less. No explanation. Less. No explanation.	
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	ode 10 11 12 19 orrec 70 71 72 73 74 75	Response Less. Mentions that (much) energy is transformed to heat. Less. Mentions that energy is needed to warm up the lamp. Less. Mentions that energy is needed to warm up the lamp. Less. Mentions that energy (heat) is lost to the surroundings. Less. Other correct. t Response The same. With erroneous explanation. Examples: Energy is always preserved. When the sun is out you don't need electrical energy. The same. No explanation is given. More. With or without explanation. Less. No explanation. Less. Energy is lost in transport. Example: Electricity is lost in the wire Less. Other erroneous explanations.	
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	D de rrect 10 11 12 19 00rrec 70 71 72 73 74 75 79 nrest	Response Less. Mentions that (much) energy is transformed to heat. Less. Mentions that energy is needed to warm up the lamp. Less. Mentions that energy is needed to warm up the lamp. Less. Mentions that energy (heat) is lost to the surroundings. Less. Other correct. t Response The same. With erroneous explanation. Examples: Energy is always preserved. When the sun is out you don't need electrical energy. The same. No explanation is given. More. With or without explanation. Less. No explanation. Less. No explanation. Less. Energy is lost in transport. Example: Electricity is lost in the wire Less. Other erroneous explanations. Other incorrect. DONSE	
	D d e rrect 10 11 12 19 orrec 70 71 72 73 74 75 79 90	Response Less. Mentions that (much) energy is transformed to heat. Less. Mentions that (much) energy is transformed to heat. Less. Mentions that energy is needed to warm up the lamp. Less. Mentions that energy is needed to warm up the lamp. Less. Mentions that energy (heat) is lost to the surroundings. Less. Other correct. t Response The same. With erroneous explanation. Examples: Energy is always preserved. When the sun is out you don't need electrical energy. The same. No explanation is given. More. With or without explanation. Less. No explanation. Less. Energy is lost in transport. Example: Electricity is lost in the wire Less. Other erroneous explanations. Other incorrect. DONSE Crossed out/erased, illegible, or impossible to interpret.	

Y2. One day when the temperature was just below 0°C, Peter and Ann made snowballs. They put a thermometer into one of the snowballs and it showed 0°C. They tried to make the snowball warmer by holding it in their hands. What do you think the thermometer showed after two minutes? Explain your answer.

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mercial purpose

Subject	ltem Key	Content Category	Performance	International Average Percent of Students Responding Correctly		International Difficulty
-			Expectation	Upper Grade	Lower Grade	Index
Science	next page	Physics	Theorizing, Analyzing, and Solving Problems	13%	10%	792

Als rentine

Y-2 Coding Guide

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One day when the temperature was just below 0°C, Peter and Ann made snowballs. They put a thermometer into one of the snowballs and it showed 0°C. They tried to make the snowball warmer by holding it in their hands. What do you think the thermometer showed after two minutes? Explain your answer.

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on purposir r Code Response Correct Response Reports 0 degrees or mentions "the same temperature". The 20 explanation includes: Snow cannot be warmer than 0 degrees. *Example:* The melting point of snow is 0 degrees. 29 Other correct. Partial Response 0 degrees or "the same temperature". No explanation or an 10 incorrect explanation. 19 0 degrees or "the same temperature" *Example:* Some snow melts, but it will not be warmer. Other partially correct. Incorrect Response 70 Above 0 degrees, because the hands are warm. 71 Above 0 degrees, because the snow melts. 72 Above 0 degrees: No explanation. 79 Other incorrect. Nonresponse 90 Crossed out/erased, illegible, or impossible to interpret. 99 BLANK

While purposes

It takes 10 painters 2 years to paint a steel bridge from one end to the other. Z1. The paint that is used lasts about 2 years, so when the painters have finished painting at one end of the bridge, they go back to the other end and start painting again.

Why MUST steel bridges be painted?

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A new paint that lasts 4 years has been developed and costs the same as the old paint. Describe 2 consequences of using the new paint. e expression

MINICOUTER

Z-1a

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Part A

b.

	Subject	ltem Key	Content Category	Performance	International Average Percent of Students Responding Correctly		International Difficulty
				Expectation	Upper Grade	Lower Grade	Index
-	Science	next page	Chemistry	Theorizing, Analyzing, and Solving Problems	63%	57%	516

Z-1a Coding Guide

Z1. It takes 10 painters 2 years to paint a steel bridge from one end to the other. The paint At tasks to paintes 2 years to pain a seer orige non-one cild to the other ine pain that is used lasts about 2 years, so when the painters have finished painting at one end of the bridge, they go back to the other end and start painting again. Why MUST steel bridges be painted? A new paint that lasts 4 years has been developed and costs the same as the old

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b. A pa	nt. Describe 2 consequences of using the new paint.
	TIMES Desulation 2 New Pool Convicted 0 4004 by ICA. The Lingue
Reproduced from	TIMSS Population 2 frem Pool. Copyright © 1994 by IEA, The Hague
	not be oser
	marcialexpression
A: Codes	for reason for painting
A: Codes	for reason for painting
A: Codes	for reason for painting Response
A: Codes	for reason for painting Response Response
A: Codes Code Correct 10	for reason for painting Response Evaluation to musting or correction
A: Codes Code Correct 10	for reason for painting Response Explicitly refers to rusting or corrosion. Other correct
A: Codes Code Correct 10 19	for reason for painting Response Explicitly refers to rusting or corrosion. Other correct.
A: Codes Code Correct 10 19 Incorrec	for reason for painting Response Explicitly refers to rusting or corrosion. Other correct. t Response
A: Codes Correct 10 19 Incorrec 70	for reason for painting Response Explicitly refers to rusting or corrosion. Other correct. t Response Mentions only aesthetic reasons. Explicitly refers to rusting or corrosion.
A: Codes Correct 10 19 Incorrec 70	for reason for painting Response Explicitly refers to rusting or corrosion. Other correct. t Response Mentions only aesthetic reasons. Examples: It looks nicer. It is uaby
A: Codes Correct 10 19 Incorrec 70	for reason for painting Response Explicitly refers to rusting or corrosion. Other correct. t Response Mentions only aesthetic reasons. Examples: It looks nicer. It is ugly.
A: Codes Correct 10 19 Incorrec 70	for reason for painting Response Explicitly refers to rusting or corrosion. Other correct. t Response Mentions only aesthetic reasons. Examples: It looks nicer. It is ugly. Cover up the rust so people won't see it. Refers to protecting or improving the bridge for reasons other
A: Codes Correct 10 19 Incorrec 70	for reason for painting Response Explicitly refers to rusting or corrosion. Other correct. t Response Mentions only aesthetic reasons. <i>Examples:</i> It looks nicer. It is ugly. Cover up the rust so people won't see it. Refers to protecting or improving the bridge for reasons other than code 10 above:
A: Codes Correct 10 19 Incorrec 70	for reason for painting Response Explicitly refers to rusting or corrosion. Other correct. t Response Mentions only aesthetic reasons. Examples: It looks nicer. It is ugly. Cover up the rust so people won't see it. Refers to protecting or improving the bridge for reasons other than code 10 above: Examples: The paint must be renewed.
A: Codes Correct 10 19 Incorrec 70 71	for reason for painting Response Explicitly refers to rusting or corrosion. Other correct. t Response Mentions only aesthetic reasons. Examples: It looks nicer. It is ugly. Cover up the rust so people won't see it. Refers to protecting or improving the bridge for reasons other than code 10 above: Examples: The paint must be renewed. It is a long time since it was painted.
A: Codes Correct 10 19 Incorrec 70 71 72	for reason for painting Response Explicitly refers to rusting or corrosion. Other correct. t Response Mentions only aesthetic reasons. Examples: It looks nicer. It is ugly. Cover up the rust so people won't see it. Refers to protecting or improving the bridge for reasons other than code 10 above: Examples: The paint must be renewed. It is a long time since it was painted. Any combination of codes 70, 71.
A: Codes Correct 10 19 Incorrec 70 71 72 73	for reason for painting Response Explicitly refers to rusting or corrosion. Other correct. t Response Mentions only aesthetic reasons. Examples: It looks nicer. It is ugly. Cover up the rust so people won't see it. Refers to protecting or improving the bridge for reasons other than code 10 above: Examples: The paint must be renewed. It is a long time since it was painted. Any combination of codes 70, 71. Challenges the information in the question.
A: Codes Correct 10 19 Incorrect 70 71 72 73	for reason for painting Response Explicitly refers to rusting or corrosion. Other correct. t Response Mentions only aesthetic reasons. Examples: It looks nicer. It is ugly. Cover up the rust so people won't see it. Refers to protecting or improving the bridge for reasons other than code 10 above: Examples: The paint must be renewed. It is a long time since it was painted. Any combination of codes 70, 71. Challenges the information in the question. Example: You don't need to paint steel bridges.
A: Codes Correct 10 19 Incorrec 70 71 72 73 79	for reason for painting Response Explicitly refers to rusting or corrosion. Other correct. t Response Mentions only aesthetic reasons. <i>Examples:</i> It looks nicer. It is ugly. Cover up the rust so people won't see it. Refers to protecting or improving the bridge for reasons other than code 10 above: Examples: The paint must be renewed. It is a long time since it was painted. Any combination of codes 70, 71. Challenges the information in the question. Example: You don't need to paint steel bridges. Other incorrect.
A: Codes Correct 10 19 Incorrec 70 71 72 73 79 Nonresi	for reason for painting Response Explicitly refers to rusting or corrosion. Other correct. t Response Mentions only aesthetic reasons. <i>Examples:</i> It looks nicer. It is ugly. Cover up the rust so people won't see it. Refers to protecting or improving the bridge for reasons other than code 10 above: <i>Examples:</i> The paint must be renewed. It is a long time since it was painted. Any combination of codes 70, 71. Challenges the information in the question. <i>Example:</i> You don't need to paint steel bridges. Other incorrect.
A: Codes Correct 10 19 Incorrect 70 71 71 72 73 79 Nonresp 90	for reason for painting Response Explicitly refers to rusting or corrosion. Other correct. t Response Mentions only aesthetic reasons. Examples: It looks nicer. It is ugly. Cover up the rust so people won't see it. Refers to protecting or improving the bridge for reasons other than code 10 above: Examples: The paint must be renewed. It is a long time since it was painted. Any combination of codes 70, 71. Challenges the information in the question. Example: You don't need to paint steel bridges. Other incorrect. Donse Crossed out/erased, illegible, or impossible to interpret.

Z1. It takes 10 painters 2 years to paint a steel bridge from one end to the other. The paint that is used lasts about 2 years, so when the painters have finished painting at one end of the bridge, they go back to the other end and start painting again.

Why MUST steel bridges be painted?

b. A new paint that lasts 4 years has been developed and costs the same as the old paint. Describe 2 consequences of using the new paint.

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Part B, First Answer

	Subject	ltem Key	Content Category	Performance	Performance International Average Percent of Students Responding Correctly		al Average Students g Correctly	International Difficulty
				Expectation	Upper Grade	Lower Grade	Index	
•	Science	next page	Chemistry	Theorizing, Analyzing, and Solving Problems	42%	35%	636	

Part B, Second Answer

	Subject	ltem Key	Content Category	Performance	International Average Percent of Students Responding Correctly		International Difficulty
				Expectation	Upper Grade	Lower Grade	Index
-	Science	next page	Chemistry	Theorizing, Analyzing, and Solving Problems	28%	22%	732

Z-1b

Z-1b Coding Guide

Z1. It takes 10 painters 2 years to paint a steel bridge from one end to the other. The paint that is used lasts about 2 years, so when the painters have finished painting at one end of the bridge, they go back to the other end and start painting again.
a. Why MUST steel bridges be painted?
b. A new paint that lasts 4 years has been developed and costs the same as the old paint. Describe 2 consequences of using the new paint.

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- B: Codes paint lasting 4 years.
- Note: Each of the two consequences must be coded separately. The same code can be used twice.

However, if the consequences described are essentially the same, the second should be coded as 79.

2°- C

Example: They don't need to go back and start again (code 11). They can wait before they start painting again (code 79).

Note: Correct responses should be plausible in the national context.

Code	Response
Correct	Response
10	Student includes the fact that there is more profit [for the painting company or the community].
	Less painters are needed. They can paint more bridges.
11	The painters don't need to paint so often or work so hard. Examples: They can wait two years before starting again. Longer vacations for the workers. They can have another job in the meantime
12	Mentions increased unemployment or lower salary for the workers.
19	Other correct. Example: Fewer problems for the traffic.
Incorrec	t Response
70	The paint will last for a longer time.
76	Merely repeats information in the stem. Examples: It will last for four years. It will cost the same
79	Other incorrect.
Nonres	ponse
<u> </u>	Crossed out/erased, illegible, or impossible to interpret.
99	BLANK



Z-2a

First Reason

Subject	Subject Item Key Content Category Performance		Internationa Percent of Respondin	International Difficulty		
			Expectation	Upper Grade	Lower Grade	Index
Science	next page	Environmental Issues and the Nature of Science	Theorizing, Analyzing, and Solving Problems	64%	59%	437

Second Reason

Subject	Item Key	Content Category	Performance	International Average Percent of Students Responding Correctly		International Difficulty
	-		Expectation	Upper Grade	Lower Grade	Index
Science	next page	Environmental Issues and the Nature of Science	Theorizing, Analyzing, and Solving Problems	43%	37%	598

Z-2 Coding Guide

A CAN

Since water is a renewable resource and so much of it falls each year, theoretically there should be enough water for everyone on Earth. Write down TWO reasons why not everyone has enough water.

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Note: Each of the two consequences must be coded separately.

The same code can be used twice.

However, if the consequences described are essentially the same, the second should be coded as 79.

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Code	Response	
Correct	Response	
10	Mentions uneven distribution of rain or other sources of water.	
	Example: Because in areas like the desert, water is not plentiful.	
11	Mentions uneven distribution of people.	
12	Mentions pollution as a cause.	
13	Mentions high(er) consumption or waste of water in some places.	
14	Mentions that evaporation is greater in some areas.	
19	Other correct: Example: Water is very plentiful, but you can't drink all kinds of water, for instance, you can't drink sea water.	
Incorrec	t Response	
79	Other incorrect.	
Nonresponse		
90	Crossed out/erased, illegible, or impossible to interpret.	
99	BLANK	



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